# PHYTOCHEMICAL AND BIOLOGICAL STUDIES ON SOME SOUTH AFRICAN PLANTS USED IN TRADITIONAL MEDICINE FOR SKIN HYPERPIGMENTATION.



## By MTANDAZO RONDO

A dissertation submitted in partial fulfillment of the requirements for the degree of Master of Sciences at the University of the Western Cape.

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### ABSTRACT

The regulation of the Western beauty care products is currently inadequate and a huge concern to consumers throughout the world, as most of the products manufactured by personal care industry are made of synthetic additives, while the traditional skin-care products are safe products made from natural elements. Usage of traditional medicinal plants is perceived as a superior alternative to achieve improvement of the gradually deteriorating standard of manufacturing beauty care products. Therefore, the present study assesses the suitability of several South African plant species conventionally used in traditional medicine for skin hyperpigmentation treatment. Ten plant species, viz. Cassine peragua, Cassipourea gummiflua, Clivia miniata, Cryptocarya myrtifolia, Gunnera perpensa, Kigelia africana, Protorhus longifolia, Rapanea melanophloeos, Rhynchosia villosa, and Senecio serratuloides were obtained after interviewing fifteen traditional healers within the Eastern Cape Province.

The methanolic extracts of the ten suggested plants were subjected to a preliminary bio-evaluation against the tyrosinase enzyme. The results showed that *Rhynchosia villosa* root was the most active extract and hence, the *R. villosa* extract was subjected to further chromatography and studied. Five known flavonoids were isolated for the first time from *R. villosa*. The isolated constituents include genistein (C1), cajanin (C2), 2'-hydoxygenistein (C3), catechin (C4), and gallocatechin (C5). The tyrosinase inhibitory activity of the isolated compounds showed the highest activity for C1 (31.45  $\mu$ M), followed by C4 (36.86  $\mu$ M), C2 (38.97  $\mu$ M), C5 (60.40  $\mu$ M), and lastly C3 (69.49  $\mu$ M).

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To the best of our knowledge, and according to an extensive search of the Sci-Finder database, the dissertation is the first scientific report to be carried out on the phytochemical and tyrosinase inhibition of *R. villosa* plant species. The *R. villosa* showed a high tyrosinase inhibitory activity, which makes it a likely candidate as a tyrosinase inhibitor that may be used as depigmenting agent for the treatment, or prevention, of pigmentation disorders in relation to hyperpigmentation. Furthermore, the results obtained also justify the traditional uses of the plant for skin depigmentation.

**Keywords**: tyrosinase enzyme, hyperpigmentation, traditional medicine, *Rhynchosia villosa*.



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### DECLARATION

I, Mtandazo Rondo, hereby declare that the dissertation entitled 'Phytochemical and biological studies on some South African plants used in traditional medicine for skin hyperpigmentation' is a bona fide record of my independent research work and has not been previously submitted for any academically purposes in any other institution. In the scope of the dissertation I have not used anybody's work as my own; the used sources have been ascribed in the correct way, thus, by means of citation, quotation, and complete reference. I further declare that ethical guidelines were complied with in conducting the study.

Signature:

Pamas

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**Date:** <u>May 2017</u>

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**Cupido** (SANBI; Kirstenbosch) for authenticating the plant species examined in the research project, and to **Dr Olugbhenga Popoola**, for technical training and support.

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The benevolence of the above-mentioned individuals made me to affirm with the words of the Brazilian novelist, Coelho Paulo, that "when you want something, all the universe conspires in helping you to achieve it." I wish that the **Mighty God** may bless all of them and their families!

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### DEDICATION

To the late Mr Zekanzima Jourbert Ndzoyiya (15 July 1949 — 3 April 2016)

A leading light traditional healer from Lusikisiki in the Eastern Cape Province (S. Africa).

He started his healing journey in year 1985.



In his lifetime as a traditional healer, he practiced in various

Southern African countries, which include Botswana, Namibia, Lesotho, and Swaziland.

He was a dignified man, with great wisdom, persistence, and compassion.

I truly believe that the study wouldn't have been feasible without him.

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### LIST OF ABBREVIATIONS

 $\lambda$  Wavelength

β Beta (equatorial position)

α Alpha (axial position)

 $\delta_C$  Carbon chemical shift

 $\delta_H$  Proton chemical shift

1D NMR One-dimensional nuclear magnetic resonance

2D NMR Two-dimensional nuclear magnetic resonance

<sup>1</sup>H NMR Proton nuclear magnetic resonance

<sup>13</sup>C NMR Carbon 13 nuclear magnetic resonance

BBO Broadband observe

CD<sub>3</sub>OD Deuterated methanol

CFR Cape floristic region

d doublet

DCM Dichloromethane

dd doublet of doublets

DEPT Distortionless enhancement by polarization transfer

DIW Deionized water

DMSO Dimethyl sulphoxide

EtOH Ethanol

EtOAc Ethyl acetate

GTPs Green tea polyphenols

Hex Hexane

HIV/AIDS Human Immunodeficiency Virus/Acquired Immunodeficiency Syndrome

HIV-OMH HIV-associated oral mucosal melanin hyperpigmentation

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HMBC Heteronuclear multiple-bond correlation

HPLC High performance liquid chromatography

HSQC Heteronuclear single-quantum coherence

Hz Hertz

IC<sub>50</sub> Half maximal inhibitory concentration

IDRC International Development Research Centre

J Coupling constant

*m* multiplet

MeOH Methanol

MHz Megahertz

min minutes

mM millimolar

NMR Nuclear magnetic resonance spectroscopy

nm nanometre

NOESY Nuclear overhauser effect spectroscopy

OHGs *ortho*-hydroxygenisteins

pH Potential hydrogen

PIH Post-inflammatory hyperpigmentation

ppm parts per million

s singlet

S/No. Structure number

SANBI South African National Biodiversity Institute

TLC Thin layer chromatography

UV Ultraviolet

viz. *videlicet* (that is to say; namely)

### LIST OF ISOLATED COMPOUNDS

The *Rhynchosia villosa* methanolic extract afforded the following five compounds:

HO 
$$\frac{8}{5}$$
 O  $\frac{1}{4}$  OH  $\frac{3}{4}$  OH  $\frac{1}{6}$  OH  $\frac$ 

**CHAPTER ONE** 

**INTRODUCTION** 

WHY NATURAL PRODUCTS STUDY IN COSMETICS?

"The perception is that natural ingredients are more pure and kinder to skin than something made in the lab, but nothing could be farther from the truth" — Dr. Joel Schlessinger

1.1 THE PROBLEM

Epidemiological data shows that about 35 % of women in South Africa use skin whitening products, which consist of synthetic chemicals, to bleach their skin (United Nation Environmental Programme, 2008). The inadequate regulation of beauty care products is a global issue. This is attested to by the dialectical claims emanating from various environment extremist groups, such as the Campaign for Safe Cosmetics and the Environmental Working Group, who proclaim that many of the cosmetics used by consumers worldwide are poisonous (Gattuso, 2011). According to their research, it has been found that a number of cosmetics consist of paraben compounds; such compounds are linked to breast cancer and hormone dysfunction (Gray, 2010; Environmental Working Group, 2016). The most common dermatological disorders in S. Africa are associated with hyperpigmentation; this is due to the fact that 91 % of the S. African population has a dark complexion. Dermatological studies conducted in Durban amongst 900 women with dark complexions showed that 67 % of the women who frequently use skin whitening products have genuine hyperpigmentation disorders (Webmaster, 2014).

1.2 SOUTH AFRICAN FLORA USE IN TRADITIONAL MEDICINE

It is clear that more work needs to be done in enhancing the standard of cosmetics. Hence, the use of traditional medicinal plants which consist of natural products is regarded as an efficient way to ameliorate the standard of manufacturing beauty care products. In the context of the

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dissertation, the phrase 'natural products', refers to substances or compounds that are produced by nature and they are directly extracted from plants (Dorland's Illustrated Medicinal Dictionary, 2000). On that note, the abundance of plant species in S. Africa is exceptional. S. Africa is one of the most plant fecund countries, consisting of approximately 23 420 plant species of about 368 families, with more than 50 % of such flora found nowhere else in the world (Cowling and Hilton-Taylor, 1994; Willis, 2006).

Nonetheless, apart from the profusion of plant species in S. Africa, knowledge regarding the phytochemistry and efficacy of such plant species remains insufficient. Thus, out of the enormous number of plant species found within the country, only 50 plant species are presently traded to any substantial level (Welford and Le Breton, 2008). This therefore necessitates more, efficient research, which will lead to innovative use of the S. African plants. The approach of using plant species in the discovery of therapeutics remains an ultimate reservoir for the effective usage of plants (Hostettmann, 1998).

Although the exploitation of plant species is quite meager when it comes to commercialization in the African continent as a whole, it is certain that the traditional medicinal practice and use of remedies made from indigenous plants play a significant role in the primary health care of millions of sub-Saharan African natives. This claim is attested to by the International Development Research Centre (IDRC), and according to the IDRC survey, about 85 % of Africans occupying the sub-Saharan African regions depend on traditional healers and their remedies for primary health care (Stanley, 2004). This claim is also supported by the large number of traditional medicinal practitioners within each sub-Saharan African country. In S. Africa alone, there are about 200 000 traditional healers (Willis, 2006).

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Furthermore, the use of traditional medicinal plants as a primary health care by a variety of culturally diverse African populations has existed for thousands of years, long before the introduction of the Western medicine (Veilleux and King, 1996). Traditional medicinal remedies are used to treat a variety of ailments and infections, which include tuberculosis, nervous complaints, depression, headache, fever, rebalancing the digestion system, and skin disorders, to mention but a few. Therefore, the current study focuses on the treatment of skin disorders, particularly the treatment of skin hyperpigmentation by the usage of medicinal plants.

### 1.3 NATURAL COSMETICS

The usage of traditional medicinal plants in curing dermatological disorders is perceived as the most efficient approach, to such an extent that beauty care manufacturers predominantly adopt this method to produce 'friendly' skin-care products (Draelos, 2008). This approach of using natural products in the manufacturing of cosmeceuticals is encouraged by the fact that, unlike synthetic drugs which are harmful to the skin, they result in improvement of the skin appearance through the delivery of nutrients which are necessary for a healthy skin (Rebeiro *et al.*, 2015). Scientific findings show that traditional medicinal plants were amongst the first beauty care products, and, interestingly, the use of natural products in manufacturing cosmeceuticals is still perceived as an efficient method, but with substantial advancements. Thus, with new formulation techniques, control of raw materials and greater standardization are achieved, the aforementioned advances have led to innovative cosmeceuticals (Kirk-Othmer, 2013).

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#### 1.4 AIM OF THE STUDY

The ultimate aim of the current study is to evaluate several plants traditionally used for the treatment of skin disorders, as tyrosinase inhibitors, and then select the most active extract for further investigation.

### 1.5 OBJECTIVES OF THE STUDY

### The purpose of the research project is to:

- ✓ Collect various plant species which are conventionally used for skin hyperpigmentation from their natural habitats, identify and document them.
- ✓ Assess the extracts of Cassine peragua (bark), Cassipourea gummiflua (bark), Clivia miniata (root rhizome), Cryptocarya myrtifolia (stem bark), Gunnera perpensa (rhizome), Kigelia africana (fruit), Protorhus longifolia (bark), Rapanea melanophloeos (bark), Rhynchosia villosa (root and leaves), and Senecio serratuloides (leaves) against tyrosinase activity.
- ✓ Use various chromatographic methods to separate bioactive compounds and to use bioassays to detect activity from the most active extract against the tyrosinase enzyme.
- ✓ Thereafter, the active compounds would be characterized, identified, and evaluated against the tyrosinase enzyme.

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### **CHAPTER TWO**

#### LITERATURE REVIEW

#### 2.1 GENERAL REVIEW

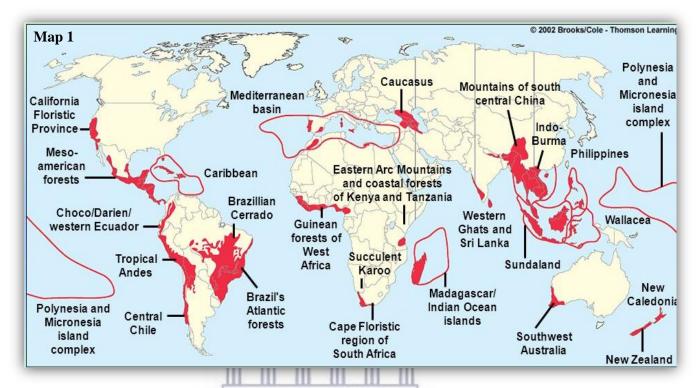
### 2.1.1 SOUTH AFRICAN FLORA

The world comprises of approximately 260 000 plant species (Miller and Spoolman, 2008). As mentioned previously, S. Africa is a country that encompasses a large number of plant species; coming sixth after Brazil (56 215), Columbia (51 220), China (32 200), Indonesia (29 375) and Mexico (26 071), with 23 420 plant species (Butler, 2016). S. Africa is the only country in the world with an entire plant kingdom within its boundaries, and around 10 % of the world's flowering plant species (angiosperms) are found in S. Africa (Fry, 2010; Germishuizen and Meyer, 2003). The world is divided into six floral kingdoms, namely, Boreal Kingdom (20 million km²), Antarctic Kingdom, Neotropical Kingdom, Australian Kingdom, Paletropical Kingdom, and the smallest, but ironically the most diverse of them all, is the Cape Floral Kingdom of the Western Cape Province of S. Africa, with only a 90 000 km² surface area (Charters and Madre, 2015).

The Cape floristic region (CFR) is the most fertile region on Earth, as it consists of the highest density of plant species globally with 31.9 % of the species being endemic (Takhtajan, 1986). This kingdom comprises roughly 8 700 plant species, with approximately 6 000 (69 %) plant species found nowhere else in the world (Goldblatt, 1978; Bond and Goldblatt, 1984). The CFR ranges from Cape Point (W. Cape Province) to Grahamstown (Eastern Cape Province) and minimally into the Northern Cape Province (Figure 2.2). Although the CFR only represents less than 0.5 % of the African continent's land surface area (Figure 2.1), it is home to nearly 20 % of

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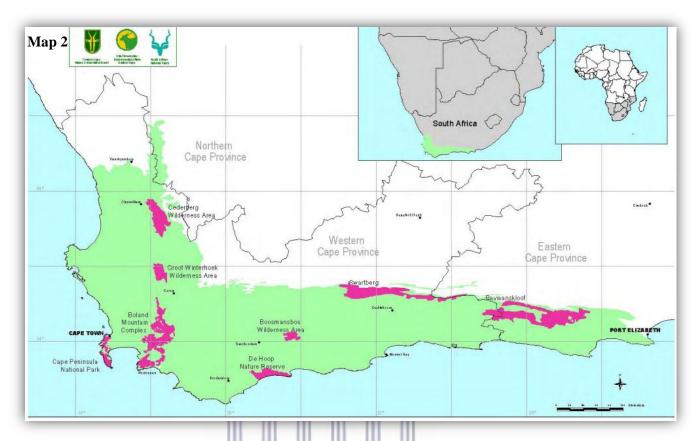
the continent's flora, making the CFR rank number seven in the biodiversity 'hotspots' of the world (Goldblatt and Manning, 2000).



Retrieved from SlidePlayer website: http://slideplayer.com/slide/8466137/

Figure 2.1: Floristic regions throughout the world.

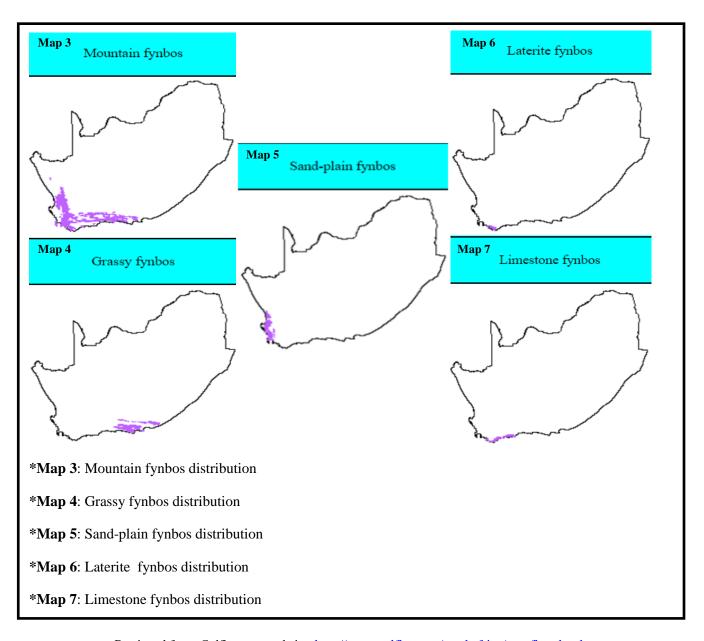
The CFR is one of S. Africa's eight World Heritage sites and is made up of eight protected areas (Figure 2.2) that are considered to be the most important examples of the CFR, namely, Cape Peninsula National Park (Table Mountain), Boland Mountain Complex, Groot Winterhoek Wilderness Area, Cederberg Wilderness Area, De Hoop Nature Reserve, Boosmansbos Wilderness Area, Swartberg Mountains, and Baviaanskloof Wilderness Area, such areas spans from the Cape Peninsula to the E. Cape Province (Anon, 1999).



Retrieved from UNESCO website: http://whc.unesco.org/uploads/nominations/1007rev.pdf

Figure 2.2: Cape floristic region.

Moreover, the primary vegetation type of the W. Cape is 'fynbos' (word derived from Afrikaans meaning 'fine bush'), and this distinctive vegetation of sclerophyllous shrubland constitute about 80 % of all the vegetation of the Cape Floral Kingdom, while 20 % is shared amongst other vegetation such as Thicket biomes, Nama Karoo and Succulent Karoo (Anon, 1999). There are five main vegetation types of fynbos; mountain fynbos being the largest, followed by grassy fynbos, sand-plain fynbos, laterite (coastal) fynbos, and limestone fynbos, and their distribution within the CFR is depicted in Figure 2.3 (Low and Rebelo, 1996).



Retrieved from Calflora.net website: <a href="http://www.calflora.net/southafrica/capeflora.html">http://www.calflora.net/southafrica/capeflora.html</a>

Figure 2.3: Main fynbos vegetation types.

Moreover, the fynbos biome is mainly made up of fine-leaved shrubs such as heaths (Ericaceae), larger-leaved showy flowered (Proteaceae), bunches of reed-like (Restionaceae), and fan-like (Iridaceae) plants (Bond and Goldblatt, 1984). The plant taxa recorded for S. Africa varies over time, due to new discoveries and taxonomic research. For instance, 268 new species of S. African plants have been described between 1994 and 2004 (Willis, 2006), hence the information in Table 2.1 below may have changed slightly as it was documented in 1984.

**Table 2.1**: The largest families and genera in the Cape Flora Kingdom

Family	<b>Total species</b>	<b>Endemic species</b>	Largest genus	Species
Daisy: Asteraceae	986	608	Senecio	133
Heath: Ericaceae	722	700	Erica	550
Vygle: Mesembryanthemaceae	660	507	Ruschia	138
Pea: Fabacea	644	525	Aspalathus	245
Iris: Iridaceae	612	485	Gladiolus	88
Protea: Proteaceae	320	306	Leucadendron	80
Cape reed: Restionaceae	310	290	Restio	85
Figwort: Scrophulariaceae	310	160	Selago	59
Buchu: Rutaceae	259	ITY 242 he	Agathosma	130
Bellflower: Campanulaceae	222	157	Lobelia	42
Orchid: Orchidaceae	206	124 E	Disa	52
Sedges: Cyperaceae	203	124	Ficinia	57
Milkwort: Polygalaceae	139	117	Muraltia	106
Jujube: Rhamnaceae	136	122	Phylica	133
Storkbill: Geraniaceae	133	67	Pelargonium	125
Sorrel: Oxalidaceae	129	90	Oxalis	129
Rose: Rosaceae	114	97	Cliffortia	106

\*Source: Bond and Goldblatt (1984)

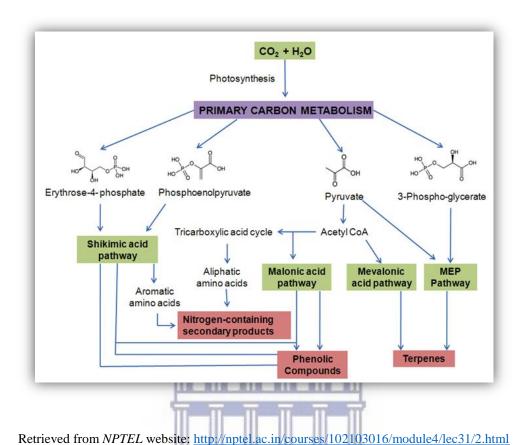
### 2.1.2 NATURAL PRODUCTS AND THEIR USE IN COSMETICS

As stated above, although there is a surplus of plant species in S. Africa, these plants have not been effectively used. It has been identified that all plant species produce natural products. These plant-derived substances, also known as 'natural ingredients' or 'secondary metabolites', can be found in different parts of the plant, i.e., the roots, flowers, stems, barks, fruits, leaves and seeds (Chugh *et al.*, 2012). Each and every plant species has a distinctive mechanism for biosynthesizing secondary metabolites, referred to as secondary metabolism (Scheme 2.1) and such mechanism is responsible for the uniqueness of a plant species (Maplestone *et al.*, 1992). It is very important to note that natural products or secondary metabolites are unnecessary for the growth of the plant species, although they contribute to its colour, aroma and flavour (Fraenkel, 1959; Saxena *et al.*, 2013). They come into existence in plant species due to the environmental conditions under which the plant germinates (Colegate and Molyneux, 2008). Natural products are also produced to protect the plant species against predators, pollution, stress, drought, ultraviolet (UV) exposure and pathogenic attacks (Schafer and Wink, 2009).

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In terms of the chemistry, the secondary metabolites are classified based on their chemical structures, composition and their solubility in various solvents or the pathway in which they are synthesized (Kabera *et al.*, 2014). The various groups of secondary metabolites include alkaloids, steroids, terpenoids, phenols, flavonoids, tannins *etc.* and they are differentiated by the functional groups present (Krishnaiah *et al.*, 2007). Such compounds are a subdivision of three chemically distinct groups, namely, nitrogen-containing secondary products, phenolic compounds and terpenes (Van Wyk and Wink, 2004).

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Scheme 2.1: Major pathways of secondary metabolites biosynthesized from primary metabolites

(Secondary metabolism).

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It has been observed that each plant species produces a characteristic mix of these bioactive compounds, which are regarded to be high value products compared to their precursors, the primary metabolites (Veberič, 2010). Secondary metabolites are considered to be fine chemicals, and are extensively used as pharmaceuticals, flavourants, pesticides, cosmetics, *etc*. (Balandrin *et al.*, 1985). The focal point of the present study is based on the latter, i.e., the use of natural products in cosmetics. The use of natural products in cosmetics has existed for centuries and has evolved drastically in recent years. As mentioned previously, natural products are perceived as components that have less adverse effects compared to synthetic drugs; thus, they are observed

as natural and free from all the harmful synthetic chemicals, with anti-tyrosinase, antioxidant and antimicrobial activities (George, 2011; Rebeiro *et al.*, 2015).

On the same note, being beautiful and feeling young are qualities that give the sense of pleasure and are, perhaps, the desire of every human being throughout the world. Hence the prevalent use of cosmetics (Kapoor, 2005). A huge problem is that many cosmetics consist of synthetic additives which are harmful or poisonous, leading to a rising number of consumers being concerned about marketed cosmetics. Thus the demand for cosmetics with traceable and more natural ingredients, free from harmful chemicals and with an emphasis on the rejuvenative properties of botanicals is instigated (Ferrer *et al.*, 2012).

It is crucial to define the term 'cosmetics' in order to avoid misperceptions. According to the European Parliament and the Council of the European Union (2009: 64), 'cosmetic' refers to:

Any substance or mixture intended to be placed in contact with the external parts of the human body (epidermis, hair system, nails, lips and external genital organs) or with the teeth and the mucous membranes of the oral cavity with a view to exclusively, or mainly, to cleaning them, perfuming them, changing their appearance, protecting them, keeping them in good condition or correcting body odours.

Also, a misconception between the term 'cosmetics' and 'cosmeceuticals' is frequently discerned. The term 'cosmeceuticals,' particularly refers to cosmetics that are known to have healing properties. Thus, it serves as a combination between drug effect and cosmetic effect; cosmeceuticals are intended to improve both the health and beauty of the skin by external application (Elsner and Maibach, 2000). The research is based on examining the tyrosinase inhibitory effect of natural products from medicinal plants which are traditionally used as cosmeceuticals.

#### 2.1.3 TYROSINASE EFFECT

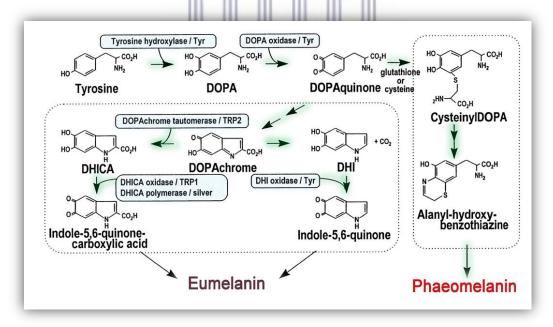
Tyrosinase, also known as polyphenol oxidase (PPO), is defined in The American Heritage Dictionary of an English Language (2016) as "any of a family of copper-containing enzymes found in animal and plant tissues, fungi and bacteria, that catalyze the oxidation of phenolic compounds and are responsible for the production of the pigment melanin from tyrosine and for the browning of fruits, vegetables, and mushrooms when cut and exposed to air". The interest of the current research is mainly centered on inhibition of the tyrosinase enzyme, to avoid extra production of the human pigment melanin. Melanin is the pigment accountable for the colour of the eyes, hair and skin; its occurrence in human skin, takes place in a cell called melanocyte via a process known as melanogenesis (Sturm *et al.*, 1998). Melanogenesis is in control of the biological reactions that defend the skin from free radical attacks which lead to cellular injury of the skin (Narayanaswamy *et al.*, 2011).

Scheme 2.2 portrays the biosynthesis of melanin (melanogenesis). In the process, tyrosinase increases the rate of the transformation reaction of L-tyrosine to L-dopa by hydroxylation and it also leads to the formation of dopaquinone from L-dopa by oxidation (Mason, 1955). A fast transformation of dopaquinone to dopachrome occurs via series of non-enzymatic reactions (Seiji, 1967). Tyrosinase appears once again in the process, catalyzing the rate of transformation from dopachrome to the formation of melanin. Dopachrome is regarded as a possible control point of melanin biosynthesis (Körner and Pawelek, 1980). However, not all substances that can inhibit the formation of dopachrome are tyrosinase inhibitors, e.g., thymol (Satooka and Kubo, 2011). Melanocyte cells are capable of producing two types of melanin pigments viz. 'eumelanin' (black or brown) and 'pheomelanin' (red or yellow) as depicted in Scheme 2.3 (Commo *et al.*, 2004).

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### Retrieved from Pepperdine Biochemistry

Scheme 2.2: Melanin biosynthesis (Melanogenesis process).



Retrieved from BIL 250 website: http://www.bio.miami.edu/dana/250/250SS15\_6.html

**Scheme 2.3**: Production of different pigments by melanocytes.

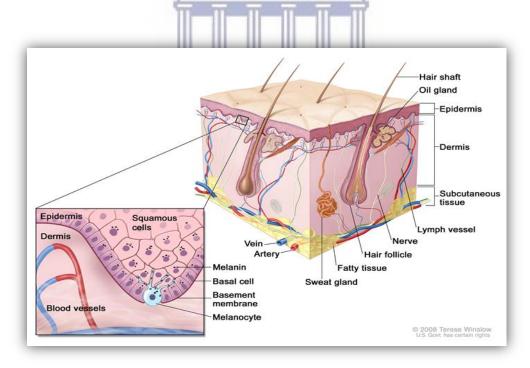
Since melanogenesis is a series of biological reactions which proceeds through a free radical mechanism, it could be disrupted by the selective use of antioxidants which are effective enough to alter these reactions (Momtaz *et al.*, 2008). The unavoidable production of free radicals, such as different reactive oxygen species (ROS) of aerobic metabolism, leads to cellular injury of the skin (Devasagayam *et al.*, 2004). These harmful oxidants are mainly due to environmental pollutants and UV radiation from sunlight and together with other free radicals damage the skin, instigating skin disorders (Silva *et al.*, 2010). ROS increase the melanin biosynthesis, which is detrimental to deoxyribonucleic acid (DNA) and they are likely to cause proliferation of melanocytes (Yasui and Sakurai, 2003).

As stated above, the enzymatic effects of tyrosinase are also found during browning of food when stored or processed in air. Such effects lead to a loss of nutritional quality, decrease in digestibility and formation of anti-nutritional and toxic compounds in the food (Friedman, 1996). Hence, tyrosinase inhibitors are also needed to inhibit browning reactions and preventing adverse effects of browning during food storage, processing and consumption (Rao *et al.*, 2013).

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### 2.1.4 SKIN DISEASE; HYPERPIGMENTATION

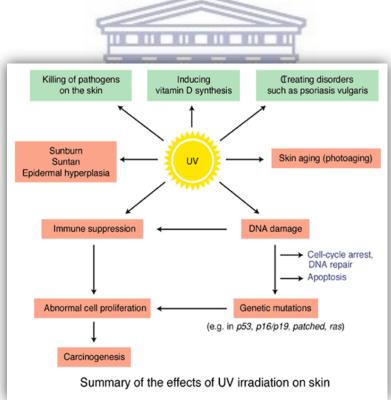
The largest and outermost organ in the human body is the skin. It has a surface area of approximately 1.8 m<sup>2</sup> and accounts for about 16 % of the body mass in the average adult (Thomas, 1993; D'Orazio *et al.*, 2013). This organ is made up of three layers (Figure 2.4), viz. the epidermis (which acts as a barrier to external influences and water loss from the inside), dermis (contains blood vessels to supply oxygen and nutrients), and subcutaneous tissue also known as hypodermis, which consists of connective tissue and fat (Weller *et al.*, 2008). Generally, the skin protects the body from the external environment, regulating temperature, aiding in fluid balance, prohibiting dangerous microbes and chemicals from the body and defending the body from the sunlight (Rebeiro *et al.*, 2015). Thus, the skin is the most vulnerable organ to invasion by bacteria, viruses, and other toxic elements (Narayanaswamy *et al.*, 2011).



Retrieved from National Cancer Institute website: http://www.cancer.gov/types/skin/hp/skin-genetics-pdq

**Figure 2.4**: Three-dimensional diagram of the skin.

It is known that 34 % of all occupational diseases are skin disorders (Spiewak, 2000). Skin disorders have a great variation, particularly when it comes to symptoms and severity, thus, they can be classified as genetic or circumstantial, permanent or temporary, lethal or harmless, painful or painless (Brind'Amour, 2016). UV radiation from the sun is seen as the major cause of skin diseases. Penetration of UV radiation leads to variations in the composition of the skin, including accumulation of the elastic fibers, collagen reduction and degeneration, and the deposition of glycosaminoglycan (Braverman and Fonferko, 1982; Olikarnin and Kallionen, 1989; Smith *et al.*, 1962). It also leads to various skin disorders (Scheme 2.4) such as, skin cancer, age spots, photosensitivity, moles, polymorphous light eruption (PMLE), solar elastosis (wrinkles), *etc.* (Kennedy, 2015).



Retrieved from Healthy Fellow website: http://www.healthyfellow.com/220/skin-aging-and-ellagic-acid/

**Scheme 2.4**: Summary of the effects of UV irradiation on skin.

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Skin darkening (i.e., hyperpigmentation) is a result of extra production of melanin, which is mainly associated with UV radiation. However, UV radiation is not the only cause of hyperpigmentation, other causes include genetic disposition, aging, hormonal influences, air pollution and skin injuries or inflammatory skin conditions (Drakaki *et al.*, 2014). Hyperpigmentation is also a side effect of certain hormone treatments, chemotherapy drugs, antibiotics, anti-seizure drugs, and other medications (Butler and Elston, 2016).

Hyperpigmentation disorders are differentiated based on the multiple mechanisms that produce various types of discolourations, spots, or marks. The different types of hyperpigmentation include ephelide (freckles), sun spots (age spots), stretch marks, moles, melanoma, post-inflammatory hyperpigmentation (PIH), melasma (chloasma), seborrheic keratoses, acanthosis nigricans, diabetic dermopathy, tinea versicolor (pityriasis versicolor), *etc.* (Just About Skin, 2014; Stulberg *et al.*, 2003). However, not all of the above-mentioned forms of hyperpigmentation are explained in the scope of the dissertation, thus, only the common forms of hyperpigmentation, particularly in S. Africa is discussed.

Melasma is one of the most common types of hyperpigmentation. It is also known as chloasma or 'pregnancy mask'. Although the latter name and the fact that an estimated 70 % of women who get affected by this hyperpigmentation during their pregnancy might suggest that melasma is only for pregnant women, this is simply not true, as studies revealed that 10 % of men have such skin discolouration too (Moin *et al.*, 2006; Vazquez *et al.*, 1988). Of course, such hyperpigmentation is common to pregnant women, particularly those with a darker complexion, but, it also affects adolescents and older women as a result of side effects of certain medications (Jadotteand Schwartz, 2010). Melasma appears as dark patches or hyperpigmented macules on the face, neck and rarely the upper limbs (Bagherani *et al.*, 2015). It is somewhat idiopathic,

especially to men, thus, its pathogenesis is unclear, but it is anticipated to be due to UV radiation, anti-seizure medications, hormonal variations of pregnancy and thyroid disease (Cayce *et al.*, 2004).

Another common type of hyperpigmentation is that of post-inflammatory hyperpigmentation (PIH). PIH is a skin condition that results from preceding cutaneous disorders, such as trauma (e.g., burns), inflammatory diseases (e.g., lichen planus, lupus erythematosus, atopic dermatitis), infections, allergic reactions, mechanical injuries, reactions to medications, and phototoxic eruptions (Schwartz and Elston, 2015). Lentigo is a common hyperpigmentation disorder; this hyperpigmentation is a circumscribed brown spot resulting from an increased number of melanocytes at the dermoepidermal junction (Ortonne, 2003). Lentigines can occur in any area of the skin, and they are classified in accordance with aetiological factors, thus, some lentigines are associated with too much exposure to UV radiation (e.g., lentigo simplex, solar lentigo, photochemotherapy (PUVA) lentigo, reticular/ink-spot lentigo, and sunbeds lentigo), some are not associated with overexposure to UV radiation (e.g., genital lentiginosis, labial lentigo, and oral lentiginosis), and other lentigines are characteristic feature of multisystem syndromes (Carli and Salvini, 2009).

Care is needed in differentiating these spots and moles, as they are subtle to those which are symptoms of Melanoma. Melanoma is the most dangerous form of skin cancer, it develops in the melanocyte, when unrepaired DNA damage due to UV radiation causes mutations (genetic defects), making the skin cells multiply rapidly, forming malignant tumours (Skin Cancer Foundation, 2016). Melanoma tumours can grow anywhere in the skin, yet they are likely to begin in the trunks (chest and back) in men and on the legs in women. They are brown or black in colour, however some can appear pink, tan, or white (American Cancer Society, 2016).

Melanoma is regarded as the most harmful skin cancer type since it is likely to spread to other parts of the body if not caught early and due to its reluctance to chemotherapy (Marieb and Hoehn, 2013).

Furthermore, pigmentation disorders in relation to hyperpigmentation are closely associated with Human Immunodeficiency Virus/Acquired Immunodeficiency Syndrome (HIV/AIDS). It has been shown that HIV infected people are more likely to have dermatological disorders; the prevalence is revealed to approach 100 % (Koehler *et al.*, 1992). The most common type of hyperpigmentation or PIH disorder affecting approximately 75 % of the people with HIV in S. Africa is that of morbilliform eruption (Jordaan, 2008). Other hyperpigmentation disorder associated with HIV-affected people is that of HIV-associated oral mucosal melanin hyperpigmentation (HIV-OMH). In S. Africa, HIV-OMH was found to be 18.5 %; relatively high when compared to other sub-Saharan African countries, e.g., Tanzania 4.7 % and Kenya 6 % (Chandran *et al.*, 2016; Chandran *et al.*, 2014). Moreover, some hyperpigmentation or PIH associated with HIV include Kaposi's sarcoma, seborrheic dermatitis, psoriasis, cutaneous vasculitis, thrombocytopenic purpura, pemphigoid, autoimmune blistering, *etc.* (Knott, 2015).

#### 2.2 REVIEW OF RHYNCHOSIA VILLOSA

#### 2.2.1 FABACEAE

Fabaceae is the predominant family of the plant order Fabales, and it constitutes nearly 95 % of this order, while 5 % is shared among Polygalaceae, Surianaceae, and Quillajaceae families (Turner, 2016). It is also one of the largest families of flowering plants (angiosperms), following Asteraceae and Orchidaceae, respectively, with about 19 400 recognized plant species of approximately 751 genera; such plant species amount to almost 7 % of angiosperms (Judd *et al.*, 2002; Christenhusz and Byng, 2016). Fabaceae is also scientifically known as Leguminosae or Papilionaceae, but, often called legume, pea or bean family (International Association for Plant Taxonomy, 2012). This family has three distinct subfamilies, namely, Faboideae, Caesalpinioideae, and Mimosoideae which have been recognized to be the only phylogenetic from this family 55–50 MYA; these subfamilies used to be treated at the family level (Herendeen *et al.*, 1992).

# 2.2.2 RHYNCHOSIA; RHYNCHOSIA VILLOSA

Rhynchosia is one of the ten genera of the subtribe Cajaninae of tribe Phaseoleae; Phaseoleae is amongst the 28 tribes of subfamily Faboideae (Moteetee et al., 2012). Rhynchosia genus has approximately 230 plant species which are prevalent the most in Africa, with approximately 60 of such species occurring in S. Africa (Schrine, 2005; Germishuizen 2006). They are also found in other continents, particularly in climates characterized by warm temperatures, such as tropical Asia, northern Australia, and also in tropical and subtropical America (Schrine, 2005). The plant species of Rhynchosia which occur in S. Africa were separated based on morphological structures into five sections by Baker in 1923; such sections include Cyanospermum (Wight and

Arnott) Benth. (1 species), *Arcyphyllum* Torrey and Gray (2 species), *Polytropia* Presl. (2 species), *Chrysoscias* E. Mey (4 species), and *Eurhynchosia* Baker (50 species) (Baker, 1923).

Rhynchosia belongs to the Eurhynchosia section and is closely related to Eriosema, which also falls within the same section, and is also one of the Cajaninae genera. The two genera have similar pinnately compound leaves and axillary racemes (Doyle and Doyle, 1993; Lackey, 1981). Such close relation between the two genera is attested to by the similarity of the morphological structure of the plant species of interest, Rhynchosia villosa, to that of Eriosema cordatum. Both plant species develop a perennial underground organ, an erect rootstock. Also, both plant species have stems and leaves with observable pubescence, and flowers of the inflorescences are yellow in colour with red veins, but the size and shape of the leaves of these two species are different, as the leaves of R. villosa adopt a peltate shape, while those of E. cordatum assume an ovate shape (Stirton and Gordon-Gray, 1978; Amrisha, 2015). However, the lack of anatomical and developmental studies on these plant species makes it problematic to precisely differentiate these plant species, as it is sometimes unclear to tell where root gives place to stem (Stirton and Gordon-Gray, 1978).

Moreover, the synonym of *R. villosa*, which is *Eriosema villosum* also substantiate the close relation between these two genera. *Eriosoma villosum* is not the only synonym of *R. villosa*; other known synonyms are *Rhynchosia sigmoides* and *Sigmodostyles villosa*. Nonetheless, *R. villosa* is endemic to S. Africa and widely distributed in KwaZulu-Natal (KZN), Mpumalanga Province, and E. Cape Province (Foden and Potter, 2005). This plant species is known as *'Uphuzi lobala'* or *'Uphuzi lentaba'* by the people of the Pondoland, its Xhosa name is derived from that of *Gunnera perpensa*, which is *'Uphuzi lomlambo*, under the misapprehension that the two plants are phylogenetical and/or morphological related.

#### 2.2.3 TRADITIONAL USES OF R. VILLOSA

The similarity of *R. villosa* (*E. villosum*) and *E. cordatum* is not merely based on their morphological structures and names, but also on their pharmacological activities. Thus, the root of *R. villosa* is posited to be traditionally used and interchangeable with that of *E. cordatum*, and other two *Eriosoma* species, *E. salignum* and *E. kraussianum* by the people of KZN for male sexual disorders, which include erectile dysfunction and/or male potency, hence this, suggests that they consist of some aphrodisiac activity (Drewes *et al.*, 2013). The above-mentioned plant species are collectively known by the people of KZN as '*Ubangalala*,' and they have been used by the S. African men or Zulu warriors for centuries to enhance their sexual energy and potency (Hutchings, 1996).

Alternatively, the *R. villosa* root is utilized by the people of the Pondoland as the main constituent in the fusion or decoction that helps women with fertility problems. Therefore, this suggests that the plant species of interest has estrogenic properties, which is attributed to its use as the ovulatory stimulant. Moreover, the people of Lusikisiki mainly use *R. villosa* root in cosmetic preparations. It is used externally and internally to enhance the skin texture and tone, while it also aids in treating swollen parts of the body when washed with water that has soaked leaves of this plant. The pharmacological and ethnopharmacological uses of some *Rhynchosia* species are presented in Table 2.2.

Table 2.2: Pharmacological and ethnopharmacological uses of some Rhynchosia species

Plant Source	Location	Pharmacological and	Reference(s)
		ethnopharmacological use(s)	
R. rothii	India	Tonic	(Gauri and Sharma, 2011)
R. rufescens	India	Abortion	(Ajesh et al., 2012)
R. beddomei	India	Antioxidant, anti-inflammatory,	(Sridharamurthy et al., 2013)
		analgesic, diuretic activity,	(Eswaraiah et al., 2013)
		and microbial activity	(Bakshu and Raju, 2001)
R. scarabaeoides	India	Antibacterial activity	(Challa et al., 2011)
R. minima	India, USA	Anthelmintic activity	(Mali and Mahale, 2008)
	and Sri Lanka		
R. volubilis	Japan	Antiproliferative activity	(Kinjo et al., 2001)
	and Korea	and used as cosmeceuticals	(Hong et al, 2016)
R. hirta	Kenya	Malaria	(Pascaline et al., 2011)
R. nulubilis	Korea	Antioxidant, (Y	'im et al., 2007; Kim et al., 2009)
		and antinociceptive	
R. molubils	Korea	Antioxidant	(Ha, 2006)
R. momons	Roica	and anti-tyrosinase activity	(1m, 2000)
		and anti-tyrosinase activity	
R. precatoria	Mexico	Tuberculosis	(Coronado-Aceves et al., 2016)
R. caribaea	Namibia	Analgesics	(Dushimemaria et al., 2015)
	1	UNIVERSITY of the	
R. capitata	Pakistan	Gastric problems flatulence,	(Wariss et al., 2014)
	1	constipation, purgative, headache	e,
		nausea and vomiting	
R. reniformis	Pakistan	Antiglycation, platelet aggregati	on, (Ahmad <i>et al.</i> , 2011)
		antioxidant, cytotoxic and phyto	otoxic
R. viscosa	Tanzania	Inflammatory skin disorders	(Crawford <i>et al.</i> , 2010)
		and insect bites	
R. recinosa	Tanzania,	Peptic ulcers, and	(Haule et al., 2012)
	and Zimbabwe	e antileishmanial activity	(Monzote <i>et al.</i> , 2014)
R. heterophylla	Zambia	Fermentation of beverage	(Zulu <i>et al.</i> , 1997)
		and anti-inflammatory	,
R. insignis	Zimbabwe	Diarrhea (Chimp	onda and Mukanganyama, 2010)

#### 2.2.4 CHEMISTRY OF RHYNCHOSIA

The chemistry of the *Rhynchosia* genus is not fully complete, since not many compounds have been identified from this genus. The insufficient information available about the chemistry and biology of this genus is due to the lack of work that has been done in examining its species in the many regions in which they occur, except in India. For instance, S. Africa has a huge number of *Rhynchosia* species (58), 57 % more than that of India (25), but information about their pharmacological activities is limited (Germishuizen, 2006; Baker, 1976; Sanjappa, 1992). As demonstrated by Tables 2.3.1–2.4.6, that show summary of the main compounds of *Rhynchosia*, intensive and novel work in this genus has been done in India by Adinarayana and core-workers. A comprehensive phylogenetic study of S. African *Rhynchosia* species was carried out by Baker in 1923. The literature available indicates that more pharmacological studies need to be done on the S. African *Rhynchosia* species.

The class of *Rhynchosia* consists of herbs, commonly climbing or trailing, sometimes woody stock, rarely shrubs (Bisby, 1994). A large number of isolated constituents from the species of this genus have been obtained from leaves. The organic compounds which are most abundant in *Rhynchosia* are flavones, mainly *C*-glycosylflavones (Rao *et al.*, 1991). Such constituents are distributed amongst edible fruits, cereals and leaves; they have important biological potential, such as anti-inflammatory, anti-diabetic, anti-spasmodic, anxiolytic, and hepatoprotective activities, hence their excellence use in human diet (Court and Williams, 2015). Also, a numerous number of compounds isolated from *Rhynchosia* are known as pro-antioxidant molecules. Such compounds include the flavones (1, 2, 6, 11, and 12), isoflavones (14 and 15), flavonols (29, 30, 32, 33, and 39), flavan-3-ol (40), flavanonols (42 and 43), phenolic acids and/or benzonoic acid derivatives (45 – 48), xanthone (53) and sterol (55) (Goufo and Trindade,

2014; Mathangi and Sudha, 2012; Banerjee *et al.*, 2012; Battu *et al.*, 2011; Rammohan *et al.*, 2015). Such compounds are found in food, suggesting that many species of *Rhynchosia* like other legumes also act as a potential source of edibles.

The above-mentioned compounds substantiate the use of some of the *Rhynchosia* species as cosmeceuticals. As in the case of *R. molubilis*, which is reported to consist of antioxidants that lead to its ability to block UV radiation, automatically prevent oxidation and remove free-radicals, it is also able to carry out skin-whitening action such as the inhibition of melanin formation (Ha, 2006). Another *Rhynchosia* species which has been found to be beneficial in depigmenting the skin and treating solar elastosis (wrinkles), is that of *R. nolubilis*, it has been revealed to be excellent in inhibiting melanin generation, while it can treat skin wrinkles through reducing the expression of collagenase MMP-1 and increasing the generation of collagen (Yim *et al.*, 2009; Hong *et al.*, 2016). On the other hand, Koreans also use *R. volubilis* as a major ingredient in local cosmetic formulations, as it is known to possess anti-inflammatory, antibacterial, anti-aging, skin whitening activities as well as prevention of wrinkles (Hong *et al.*, 2016; Lai *et al.*, 2012).

Generally, the extensive occurrence of the different classes of flavonoids (Table 2.3.1–2.3.6) in *Rhynchosia* supports the significance of this genus in cosmetic and pharmaceutical fields as anti-inflammatory and antioxidant agents (Malinowska, 2013). Gallic acid (45) and the pyrogallol-type tannins (41, 49, 51, 52, and 58) isolated from *R. volubilis* showed high antiproliferative activities against murine melanoma (B16F10) cells, hence this also promotes the cosmeceutical effects of the *Rhynchosia* species, i.e., *R. volubilis*, *R. nolubilis* and *R. molubilis* in particular (Kinjo *et al.*, 2001).

## **Group 1: Flavonoids**;

Table 2.3.1: Flavones and flavone glycosides

S/No.	Chemical structure and name of isolated	Plant	Biological	Reference(s)
	compound	source(s)	activity	
1		R. suaveolens		(Rammohan et al., 2015)
	ОН	R. heynei	Radical scavenging	(Adinarayana et al., 1985)
	НО	R. capitate R. beddomei	activity	(Rao and Gunasekar, 1987)
	Glu	R. rufescens		(Adinarayana et al., 1980c)
	OH O	R. sericea		(Adinarayana et al., 1979)
	Isovitexin	R. rothii		(Admarayana et at., 1979)
		R. cana		(Adinarayana and Ramachandraiah, 1986)
		R. jacobii		Tumachanararan, 1200)
		R. bracteata		(Seetharamamma et al., 1989)
		R. minima	2	(Nia et al., 1992)
	THE RESERVE OF THE PERSON OF T		T T	(Besson et al., 1977)
2	T T T	R. suaveolens	7	(Rammohan et al., 2015)
	ОН	R. rufescens	Radical scavenging	(Adinarayana et al, 1985)
	но	R. beddomei R. rothii	activity	(Adinarayana et al., 1979)
	Glu UNIVER	R. capitate	he	(Adinarayana et al., 1980c)
	Isoorientin	R. cana R. jacobii	E	(Adinarayama and Ramachandraiah, 1986)
		17.32		(Adinarayana and Chetty, 1985)
				(Rao and Gunasekar, 1987)
				(Seetharamamma et al.,1989)
3	OH	R. cana		(Adinarayana and
	Glu	R. jacobii		Ramachandraiah, 1986)
	но	R. heynei		(Seetharamamma et al., 1989)
		R. rufescens		(Adinarayana et al., 1979)
	Ĭ, Ĭ	R. beddomei		(A 1'
	OH O	R. hirsuta		(Adinarayana et al., 1980c)
	Orientin	R. rothii		(Adinarayana et al., 1985)
				(Rao and Gunasekar, 1987)

	<b></b>	D 4.11	1	(D 1 (1007)
4	OH OH	R. rothii		(Rao and Gunasekar, 1987)
	HO,,,	R. heynei	Anti-	(Adinarayana and
	HOOOH	R. hirsuta	inflammatory	Ramachandraiah, 1986)
	HO, O,	R. rufescens		(Adimorayana et el 1005)
		R. beddomei		(Adinarayana <i>et al.</i> , 1985)
	HO ", O	R. cana		(Adinarayana et al., 1979)
	но он о	R. jacobii		(Adinarayana et al., 1980c)
	ŌН	R. minima		(Seetharamamma et al., 1989)
	Vicenin-2			(Besson et al., 1977)
5	OH OH	R. cana		(Adinarayana and
	HO,, OH	R. beddomei		Ramachandraiah, 1986)
	J OH	R. rufescens		(Adinarayana et al., 1980c)
	HO O	R. rothii		(Adinarayana <i>et al.</i> , 1980b)
				(A.1'
	HO " O			(Adinarayana <i>et al.</i> , 1979)
	но он о		T .	(Adinarayana et al., 1985)
	ŌН		ri e	(Rao and Gunasekar, 1987)
	Lucenin-2			(rao ana Ganasekar, 1907)
6	OH	R. cana		(Adinarayana and
	Glu HO O	R. capitate		Ramachandraiah, 1986)
	HO VO	R. heynei	7	(Adinarayana et al., 1985)
	UNIVER	R. rufescens	the	(Famarayana et att., 1905)
	о́н о́ WESTE	R. beddomei	E	(Adinarayana et al., 1979)
	Vitexin	Statement Head and a	100000	(Adinarayana et al., 1980c)
7	<b>ү</b> н он			(Seetharamamma et al., 1989)
	HO,,,	R. jacobii		(Seemaramamma et at., 1989)
	но			
	HO, O,			
	HO OH Ö			
	ÖH			
	Vicenin-1			

8	HO, OH HO OH OH OH OH Vicenin-3	R. minima	(Besson et al., 1977)
9	HO, OH	R. minima	(Besson et al., 1977)
10	HO, OH OH OH OH OH	R. beddomei SITY of the	(Adinarayana <i>et al.</i> , 1980c) (Adinarayana <i>et al.</i> , 1985)
11	OH OH OH OH O Luteolin	R. suaveolens	(Adinarayana et al., 1985)

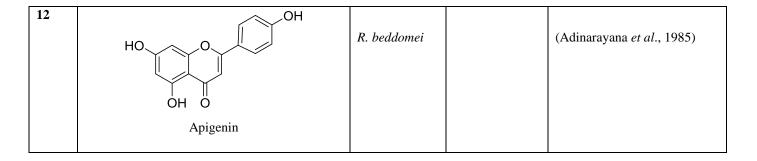


Table 2.3.2: Isoflavones

S/No.	Chemical structure and name of isolated	Plant source(s)	Biological	Reference(s)
	compound		activity	
13	OHOOHOOH Cajanin	R. edulis R. densiflora R. carabaea R. hirsuta		(Ogungbe et al., 2011) (Ingham, 1990)
14	HO O OH OH Genistein	R. densiflora R. carabaea R. hirsuta		(Ingham, 1990)
15	HO OH OH OH OH 2'-hydoxygenistein	R. densiflora R. carabaea R. hirsuta		(Ingham, 1990)
16	OH O Ulexin B	R. edulis		(Ogungbe et al., 2011)

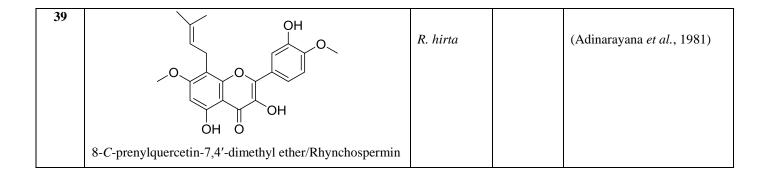
15	\ _	Т	
17	ОНООНОН	R. edulis	(Ogungbe et al., 2011)
	Scandenal		
18		R. edulis	(Ogungbe et al., 2011)
	Rhynedulin A		
19	HO	R. edulis	(Ogungbe et al., 2011)
	Rhynedulin B		
20	OH O OH	R. edulis	(Ogungbe et al., 2011)
21	OH OCyclochandalone	R. edulis of the	(Ogungbe et al., 2011)
22	OH O OH OH Rhynedulina	R. edulis	(Ogungbe et al., 2011)

22	\		
23	ОНООН	R. edulis	(Ogungbe et al., 2011)
	Caianana		
24	Cajanone  HO O		
27	ОН	R. volubilis	(Li and Xiang, 2011)
	Daidzein		
25	НО	R. volubilis	(Li and Xiang, 2011)
	Ö		
26	HO O O O	R. volubilis	(Li and Xiang, 2011)
	Biochanin A	ITY of the	
25	WESTER	N CAPE	
27	HO OH OH	R. volubilis	(Li and Xiang, 2011)
	Pratensein		
28	HO OH O	R. volubilis	(Li and Xiang, 2011)
	3-[4'-(1'-methoxy)-propoxy]-phenyl-5,7-		
	dihydroxyisoflavone		

Table 2.3.3: Flavonols

S/No.	Chemical structure and name of isolated compound	Plant	Biological	Reference(s)
		source(s)	activity	
29	OH OH HO O-Rutinose OH O Rutin/Rutoside	R. beddomei R. hirta		(Adinarayana <i>et al.</i> , 1985) (Adinarayana <i>et al.</i> , 1980c) (Adinarayana <i>et al.</i> , 1980a)
30	HO O O O Rutinose OH O Kaempferol-3- <i>O</i> -rutinoside/Nicotiforin	R. hirta		(Adinarayana and Chetty, 1985) (Adinarayana <i>et al.</i> , 1980a)
31	OII O	R. beddomei		(Adinarayana et al., 1980b)
32	HO OH OH OH OH OKaempferol	R. rufescens		(Adinarayana and Ramachandraiah, 1984)
33	HO OH OH OH OH 3-O-methylquercetin	R. rufescens		(Adinarayana and Ramachandraiah, 1984)

34	HO OH O	R. beddomei	(Adinarayana et al., 1980b)
35	HO OH OH OH 5-deoxyquercetagetin/Rhynchosin	R. beddomei	(Adinarayana et al., 1980b)
36	OH OH OH OH O Fisetin	R. beddomei	(Adinarayana et al., 1980b)
37	HO OH OH OH OH Robinetin	R. beddomei	(Adinarayana et al., 1980b)
38	HO OH O	R. beddomei	(Adinarayana et al., 1980b)



**Table 2.3.4**: <u>Flavan-3-ols</u>

S/No.	Chemical structure and name of isolated	Plant	Biological	Reference(s)
	compound	source(s)	activity	
40	HO OH OH OH OH Epicatechin	R. volubilis	?	(Li and Xiang, 2011)
41	HO OH OH OH OH OH OH 7-O-galloylcatechin	R. volubilis	Antiproliferative activity	(Kinjo <i>et al.</i> , 2001)

Table 2.3.5: Flavanonols

S/No.	Chemical structure and name of	Plant	Biological	Reference(s)
	isolatedcompound	source(s)	activity	
42	OH OH	R. densiflora R. hirta		(Rao and Gunasekar, 1998) (Rao and Gunasekar, 1988)
	6-C-prenyltaxifolin-7,3'-dimethylether/Isotirumalin			
43	OH OH OH OH OH OH OH 8-C-prenyltaxifolin-7,4'-dimethyl ether/Tirumalin	R. hirta		(Adinarayana <i>et al.</i> , 1985) (Adinarayana <i>et al.</i> , 1980a)

Table 2.3.6: Flavanone(s)

S/No.	Chemical structure and name of	Plant source(s)	Biological	Reference(s)
	isolatedcompound		activity	
44	HO OH OO Naringenin	R. beddomei		(Adinarayana <i>et al.</i> , 1980c)

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## **Group 2: Miscellaneous**;

Table 2.4.1: Phenolic acids

S/No.	Chemical structure and name isolated	Plant	Biological	Reference(s)
	compound	source(s)	activity	
45	HO OH OH Gallic aid	R. volubilis R. minima	Antiproliferative activity	(Kinjo <i>et al.</i> , 2001)  (Krishnamurty <i>et al.</i> , 1975)
46	OOH	R. hirta		(Adinarayana et al., 1985)
	OH  4-hydroxy-3-methoxybanzoic acid/Vanillic acid		2	
47	O OH HO O	R. suaveolens	Antibacterial	(Khan and Shoeb, 1984)
	3-hydroxy-5-methoxy-4-(3-methyl-2-butenyl)-2- biphenylbenzoic acid	N CAPI	Ξ	
48	OH Protocatechuic acid	R. minima		(Krishnamurty <i>et al.</i> , 1975)
49	OH OH OH OH	R. volubilis	Antiproliferative activity	(Kinjo et al., 2001)
	Glucogallin			

50	HO OH OH Gallic acid ethyl ester	R. minima		(Krishnamurty et al., 1975)
51	HO OH Gallic acid methyl ester	R. volubilis	Antiproliferative activity	(Kinjo et al., 2001)
52	OH OH OH OH OH OH OH Trigallic acid	R. volubilis	Antiproliferative activity	(Kinjo et al., 2001)

Table 2.4.2: Xanthones

S/No.	Chemical structure and name of isolated compound	Plant source(s)	Biological activity	Reference(s)
53	HO OH OH OH OH OH Mangiferin	R. suaveolens	Radical scavenging activity	(Rammohan <i>et al.</i> , 2015) (Adinarayana and Chetty, 1985)

54	ŌН ÓН		
	HO	R. suaveolens	(Adinarayana and Chetty, 1985)
	HO,,,,		
	HOOOOH		
	ОНО		
	Isomangiferin		

Table 2.4.3: Sterols

S/No.	Chemical structure and name of isolated compound	Plant source(s)	Biological activity	Reference(s)
55	$\beta$ -sitosterol	R. minima		(Ahmed <i>et al.</i> , 1992) (Krishnamurty <i>et al.</i> , 1975)
56	Galactose Stigmasteryl galactoside	ITY of the		(Ahmed et al., 1992)
57	HO	R. minima		(Ahmed et al., 1992)

**Table 2.4.4**: Galloyl derivative(s)

S/No.	Chemical structure and name of isolated	Plant source(s)	Biological activity	Reference(s)
	compound			
58	OH HO OH OH OH OH 1,6-digalloly glucose	R. volubilis	Antiproliferative activity	(Kinjo et al., 2001)

 Table 2.4.5:
 Simple aromatic natural products

S/No.	Chemical structure and name of isolated	Plant	Biological	Reference(s)
	compound	source(s)	activity	
59	OH OUNIVERS	R. suaveolens	Antibacterial	(Khan and Shoeb, 1984)
	3-hydroxy-5-methoxy-4-(3-methyl-2- butenyl)biphenyl	N CAPE		
60	O OH O	R. suaveolens		(Rammohan et al., 2015)
	2-hydroxy-3,4- dimethoxybenzophenone			
61	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	R. minima		(Krishnamurty et al., 1975)
	Hydroquinone diacetate			

Table 2.4.6: Simple non-aromatic natural products

S/No.	Chemical structure and name of isolated	Plant source(s)	Biological	Reference(s)
	compound		activity	
62	OH HO'''OH OD-pinitol	R. beddomei R. hirta		(Adinarayana et al., 1979)  (Adinarayana et al., 1980a)
63	HO OH OH OH OH OH	R. spp. R. beddomei		(Allen and Allen, 1981) (Adinarayana <i>et al.</i> , 1980c)

\*R: Rhynchosia, Glu: glucose, R. hirta = R. cyanosperma, spp: unspecified species



#### **CHAPTER THREE**

COLLECTION/EXTRACTION AND CHEMICAL CHARACTERIZATION
OF BIOACTIVE COMPOUNDS FROM TRADITIONALLY USED HERBAL
MEDICINE FOR TREATMENT OF HYPERPIGMENTATION

#### 3.1 METHODS AND MATERIALS

#### 3.1.1 CHEMICAL REAGENTS

Methanol (MeOH), ethyl acetate (EtOAc), ethanol (EtOH), hexane (Hex), dichloromethane (DCM), and dimethyl sulfoxide (DMSO) were secured from Merck (S. Africa). The mushroom tyrosinase (EC 1.14.18.1) 5771 Unit/mg, L-tyrosine, kojic acid and deuterated methanol (CD<sub>3</sub>OD) were purchased from Sigma Aldrich (S. Africa).

#### 3.1.2 THIN LAYER CHROMATOGRAPHY (TLC)

Pre-coated TLC plates of silica gel 60  $F_{254}$  [Merck (Darmstadt; Germany)] were used for TLC analysis using a UV lamp of  $\lambda_{254}/\lambda_{366}$  nm [CAMAG (Muttenz; Switzerland)] to visualize the 'spots' or bands after developing the TLC plates in various solvent systems. Chemical detection was performed after spraying with vanillin/ $H_2SO_4$  reagent and heating at 110 °C.

#### 3.1.3 COLUMN CHROMATOGRAPHY (CC)

Silica gel 60 (0.040 – 0.063 mm) (Merck Millipore; S. Africa) and sephadex LH-20 (Sigma Aldrich; S. Africa) were used in various sizes of glass columns.

#### 3.1.4 HIGH PERFOMANCE LIQUID CHROMATOGRAPHY (HPLC)

For semi-prep HPLC purification, a HPLC '1200 series' (Agilent Technologies) comprising a variable wavelength UV/Vis detector, manual injector, quaternary pump (G1311A), vacuum degasser (G1322A), column compartment (G1316A) and reversed phase  $C_{18}$  column SUPELCO (25 × 2.1 cm, 5  $\mu$ m) was used. The used mobile phase was MeOH-deionized H<sub>2</sub>O (DIW) (55:45 to 70 % in 30 min, then to 100 % in 5 min) and the flow rate employed was 1.0 mL/min.

#### 3.1.5 NUCLEAR MAGENETIC RESONANCE SPECTROSOPY (NMR)

Deuterated methanol (CD<sub>3</sub>OD) was used to prepare the samples for NMR. NMR spectra were recorded at 25 °C on a Bruker Avance III HD 400 MHz NMR spectrometer (Germany) using a 5 mm BBO probe. Standard 1D and 2D NMR pulse sequences were used to acquire 1D and 2D data respectively. The  $^{1}$ H ( $\delta_{H}$ ) and  $^{13}$ C ( $\delta_{C}$ ) chemical shifts were measured comparatively to CD<sub>3</sub>OD signals at  $\delta$  3.23, 4.78 ( $^{1}$ H NMR) and  $\delta$  48.3 ( $^{13}$ C NMR) in ppm.

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#### 3.1.6 TYROSINASE ENZYME INHIBITION

The skin enzymatic inhibitory assay was executed in the course of the study following the approach used by Vanni *et al.* (1990), with slight modification (Popoola *et al.*, 2015). Initially, a stock solution of 10 mg/mL was prepared by dissolving the plant extracts to be tested in dimethyl sulfoxide (DMSO), then 5 dilutions were performed using 50 mM sodium phosphate buffer (pH 6.5) for all working solutions to obtain the working concentrations of 100.00, 50.00, 25.00, 12.50, and 6.25 μg/mL, respectively. Kojic acid was used as a positive control in this assay. Each sample working solution of 70 μL was combined with 30 μL of tyrosinase [from mushroom, 500]

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Unit/mL in sodium phosphate buffer (pH 6.5)] in triplicate, in a 96-well plate. After incubation at room temperature for 5 min, 110 µL of the substrate (2 mM L-tyrosine) was added to each well. Final concentrations of the positive control, crude extracts and/or isolated compounds ranged from 1.0 - 100 µg/mL. The sample control was made up of each sample using sodium phosphate buffer (pH 6.5) only. The enzymatic activity was determined by measuring the absorbance at 490 nm using AccuReader M965 microplate reader (Metertech; Taiwan). The measurements were recorded immediately after incubation was carried out at room temperature (25°C) for approximately 30 min. Equation 1 was employed in determining the percentage of tyrosinase inhibition.

#### Equation 1: Percentage of tyrosinase inhibition;

% of tyrosinase inhibition= 
$$\frac{(A_{control} - A_{blank 1}) - (A_{sample} - A_{blank 2})}{(A_{control} - A_{blank 1})} \times 100 \%$$

Where  $A_{control}$  is the absorbance of the control with the enzyme,  $A_{blank \, 1}$  is the absorbance of the control without the enzyme,  $A_{sample}$  is the absorbance of the test sample with the enzyme and  $A_{blank \, 2}$  is the absorbance of the test sample without the enzyme.

#### 3.1.7 TLC BIOAUTOGRAPHY

TLC bioautograpy was carried out as described by Wangthong *et al.* (2007), with slight changes. Initially, the tyrosinase solution was prepared by dissolving 1 mL of the 1000 Unit enzyme in 1 mL of 50 mM sodium phosphate buffer (pH 6.5). The substrate was also prepared by dissolving 0.0036 g of L-tyrosine in 10 mL of 50 mM sodium phosphate buffer (pH 6.5). Then, the positive control, crude extracts and/or main fractions were dissolved in methanol to prepare a stock

solution of 1 mg/mL. The kojic acid, crude extracts and/or main fractions solutions were then spotted on the TLC plate and eluted with DCM-MeOH [(95:5) and/or (90:10)]. The developed plate was then sprayed with tyrosinase enzyme and incubated at a room temperature for 5 min. After incubation, it was sprayed with L-tyrosine (20 mM) and once again incubated at room temperature for 30 min. Indication of the presence of tyrosinase inhibitory activity amongst the investigated samples was shown by clear white zones as the background of the plate assumed a purple-greyish colour.

#### 3.1.8 STATISTICAL ANALYSIS

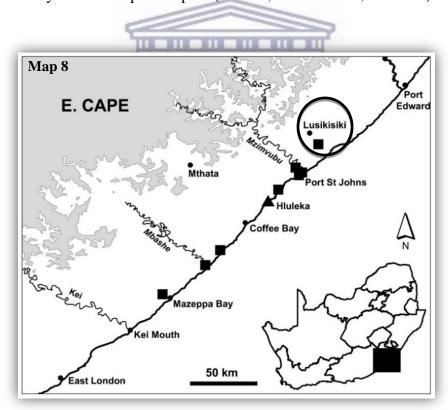
The tyrosinase enzyme inhibitory effect of the plant extracts, isolated compounds and kojic acid were assayed in triplicate. The final values of the tyrosinase skin enzymatic inhibition assay were expressed as  $IC_{50}$  values and determined using the GraphPad Prism 5.0 software. For all analyses, p < 0.05 was considered to indicate a statistically significant difference.

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#### 3.2 ACQUISITION AND AUTHENTICATION OF PLANT MATERIALS

#### 3.2.1 COLLECTION OF PLANT MATERIALS

The collection of the plant materials examined in the study was done by the late Mr Zekanzima Ndzoyiya and Mr Solomon Kuca from Ntsubane forest complex in Lusikisiki (E. Cape Province), during November 2015. Lusikisiki is located approximately 40 km north of Port St. Johns (Figure 3.1); it is characterized by spectacular views of natural vegetation and waterfalls (South African Traveler, 2015). The Ntsubane forest complex is a perpetual evergreen forest, situated on the periphery of the Indian Ocean. The voucher specimens examined in the study were authenticated by Dr. Christopher Cupido (SANBI; Kirstenbosch, S. Africa).



Retrived from Zoologische mededelingen wbsite: http://www.zoologischemededelingen.nl/83/nr03/a06

Figure 3.1: Location of Lusikisiki in the E. Cape Province (S. Africa).

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#### 3.2.2 INVESTIGATED PLANT SPECIES AND RELATIVE INFORMATION

Fifteen interviews were conducted with fifteen traditional healers within the E. Cape Province, particularly in the Pondoland (Mthatha, Libode, Port St. Johns, and Lusikisiki), to obtain ethnomedicinal knowledge and relative importance of various indigenous medicinal plants which are used in healing skin disorders and hyperpigmentation disorders, in particular. Ten plant species from various families were commonly suggested by the interviewed traditional healers. Such plant species and the relative information are shown in Table 3.1.



Table 3.1: Investigated plant species and their relative information

No.	Scientific name	Family	Herbarium number	Local name	English name	Part used	Endemism	Traditional uses	Reference(s)
1	Cassine peragua L.	Celastraceae	NBGSLD0001440	Umbomvane	Cape saffron	Bark	Endemic to Southern Africa	Traditionally used for treating burns.	_
2	Cassipourea gummiflua Tul.	Rhizophoracea	NH0006334-0	Umemezi	Large-leaved onionwood	Bark	Not endemic to South Africa	Is used as a substitute of Cassipourea malosana as skin lightener, and to treat skin ailments and sunburn. Also used for protection from evil spirits.	(Brink, 2006)
3	Clivia miniata (Lindl.) Bosse	Amaryllidaceae	PREART0003332	Umayime UNIV WEST	Bush lily ERSITY	Root rhizome of the	Endemic to Southern Africa	The rhizome is taken for snake- bite, wounds and pain relief. While whole plant is used to help with childbirth and to hasten parturition.	(Tait, 2001)
4	Cryptocarya myrtifolia Stapf	Lauraceae	SAM0043557-0	Umthungwa	Camphor tree	Stem bark	Endemic to South Africa	Used interchangeable with the bark of <i>Ocotea bullata</i> internally as in form of tea or applied as steam (external use) to treat pimples.	(Hutchings et al., 1996) (Van Wyk and Van Wyk, 1997)
5	Gunnera perpensa L.	Gunneraceae	K000350081	Uphuzi lomlambo	River pumpkin	Rhizome	Not endemic to South Africa	Externally used as a wound dressing. Infusions may be taken internally to treat psoriasis.	(Mabona and van Vuuren, 2013)

6	Kigelia africana (Lam.) Benth.	Bignoniaceae	PRE0181079-0	Umfongothi	Sausage tree	Fruit	Not endemic to South Africa	Used for various skin disorders from fungal infections, boils, acne and psoriasis, dressing wounds and sores.	(Saini <i>et al.</i> , 2009)
7	Protorhus longifolia (Bernh.) Engl.	Anacardiaceae	BNRH0010636	Uzintlwa	Red beech	Bark	Not endemic to South Africa	Used for blood-clotting related diseases.	(Mosa <i>et al.</i> , 2011)
8	Rapanea melanophloeos(L .) Mez	Myrsinaceae	PRE0822924-0	Maphipha	Cape beech	Bark	Endemic to Southern Africa	The plant species is mainly used as a charm to protect against evil spirits. Its powder form bark is internally used (chewed) and/or externally used to treat sore throats and wounds.	(Van Wyk <i>et al.</i> , 1997)
9	Rhynchosia villosa (Meissner) Druce	Fabaceae	PREART0003226	Uphuzi lobala UNIV WEST	ERSITY	Root Y of the	Endemic to South Africa	It is used externally and internally to enhance the skin texture and tone, while it also aid in treating swollen parts of the body when washed with its soaked leaves.  Also used in hormone-dependent diseases.	_
10	Senecio serratuloides DC.	Asteraceae	NHSLD0002401	Ntsukumbini	Two-day cure	Leaves	Not endemic to South Africa	Tea made from the leaves is taken in case of infections; leaves are applied directly to purulent sores. Leaf decoction is taken as a blood purifier in case of skin eruptions, while powdered leaves or roots are applied to burns or sores.	(De Wet <i>et al.</i> , 2013)  (Kelmanso <i>et al.</i> , 2000)

<sup>\*</sup>The plants were collected from Lusikisiki by Mr Z. Ndzoyiya and Mr S. Kuca, except *Kigelia african* (10) which was collected from Eerste Rivier by Ms Luveni Sonka.

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#### 3.3 INVESTIGATION OF TYROSINASE INHIBITION OF CRUDE EXTRACTS

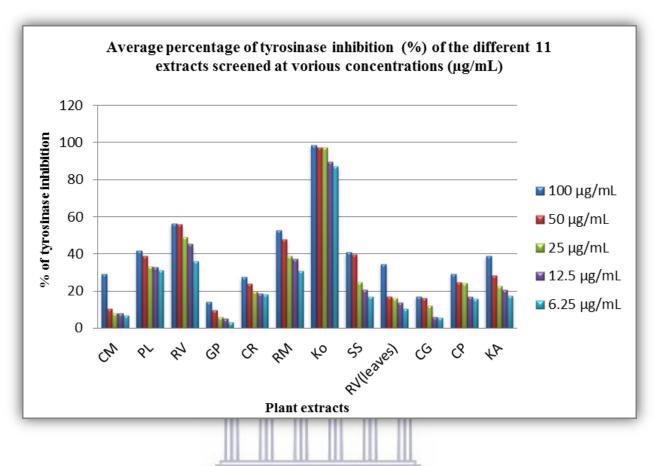
#### 3.3.1 TYROSINASE ENZYME INHIBITION OF CRUDE EXTRACTS

The tyrosinase enzyme inhibition assay was used for the preliminary bio-evaluation of the crude extracts and was carried out as described in Section 3.1.6. Kojic acid was used as a positive control in the assay since the inhibition exerted by kojic acid is well established due to its ability to chelate to copper in the active sites of the enzyme (Kubo *et al.*, 2000). Table 3.2 and Figure 3.2 show the average percentages of tyrosinase inhibition of the investigated plant species screened at various concentrations.

**Table 3.2**: Average percentages of tyrosinase inhibition (%) and detectable anti-tyrosinase  $IC_{50}$  values of the different plant extracts

	Plant		Average percentages of tyrosinase inhibition (%) in various concentration ( $\mu$ g/mL) and detectable anti-tyrosinase IC <sub>50</sub> value of the plant extract						
No.	extract code	Plant extract name	100 μg/mL	50 μg/mL	<b>25</b> μg/mL	12.5 μg/mL	6.25 μg/mL	IC <sub>50</sub> (μg/mL)	
1	CM	Clivia miniata	29.00	10.27	8.14	7.84	6.93	NA	
2	PL	Protorhus longifolia	41.63	38.96	33.33	32.88	31.28	NA	
3	RV	Rhynchosia villosa (roots)	56.40	55.78	49.16	45.28	36.00	15.67	
4	GP	Gunnera perpensa	14.31	9.59	6.09	5.18	3.20	NA	
5	CR	Cryptocarya myrtifolia	27.40	23.82	19.63	18.65	18.26	NA	
6	RM	Rapanea melanoploes	52.51	47.79	38.89	37.21	30.82	26.30	
7	Ko	Kojic acid	98.47	97.40	97.33	89.48	86.98	0.9	
8	SS	Senecio serratuloides	40.87	39.80	24.73	20.70	16.82	NA	
9	RV <sub>leaves</sub>	Rhynchosia villosa (leaves)	34.55	17.12	16.02	13.85	10.27	NA	
10	CG	Cassipourea gummiflua	17.05	16.29	12.18	6.01	5.72	NA	
11	CP	Cassine peragua	29.30	24.80	24.12	16.82	15.91	NA	
12	KA	Kigelia africana	39.04	28.39	22.83	20.47	17.50	NA	

\*NA: Not active in tested concentation range

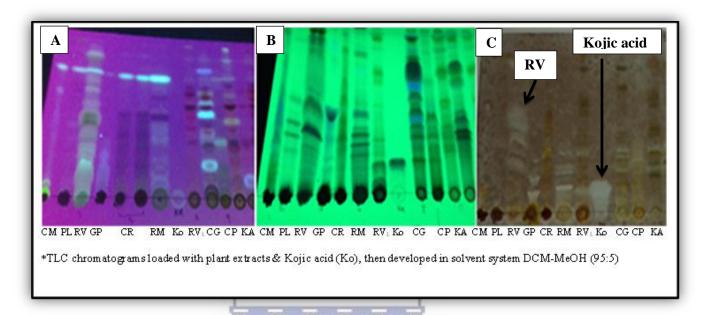


**Figure 3.2**: Bar graph depicting average percentages of tyrosinase inhibition (%) of *Clivia miniata* (CM), *Protorhus longifolia* (PL), *Rhynchosia villosa* root (RV), *Gunnera perpensa* (GP), *Cryptocarya myrtifolia* (CR), *Rapanea melanophloeos* (RM), kojic acid (Ko), *Senecio serratuloides* (SS), *Rhynchosia villosa* leaves [RV(leaves)], *Cassipourea gummiflua* (CG), *Cassine peragua* (CP), and *Kigelia africana* (KA) extracts screened at various concentrations (μg/mL).

#### 3.3.2 TLC BIOAUTOGRAPHY OF CRUDE EXTRACTS

To confirm the results obtained in Section 3.3.1, a qualitative approach was performed, i.e., that of TLC bioautography. This is a method that combines chromatographic separation and *in situ* biological activity determination. It was employed in the preliminary screening of natural products which are anticipated to have the tyrosinase enzyme inhibitory effect and for the

bioactivity-directed fractionation and isolation of active constituents from complex fractions (Cheng and Wu, 2013). The assay was executed as described in Section 3.1.7.



**Figure 3.3**: TLC chromatograms of *Clivia miniata* (CM), *Protorhus longifolia* (PL), *Rhynchosia villosa* root (RV), *Gunnera perpensa* (GP), *Cryptocarya myrtifolia* (CR), *Rapanea melanophloeos* (RM), *Rhynchosia villosa* leaves (RV<sub>L</sub>), kojic acid (Ko), *Cassipourea gummiflua* (CG), *Cassine peragua* (CP), and *Kigelia africana* (KA) extracts developed in solvent system DCM-MeOH (95:5), **A**: viewed under UV (366 nm), **B**: after spraying with vanillin/H<sub>2</sub>SO<sub>4</sub>, heating at 110°C and viewed under UV (254 nm), **C**: after spraying with tyrosinase enzyme and tyrosine.

According to Figure 3.3, it became apparent that the plant extract of RV (*R. villosa* root) was the most active, since it had clearly developed white zones of tyrosinase inhibition compared to other extracts, after spraying with both tyrosinase enzyme and tyrosine (**C**), respectively. This was the indication of the presence of bioactive constituents that are capable of inhibiting tyrosinase enzyme. Hence, *R. villosa* was chosen for further chromatographic study to isolate bioactive compounds.

# 3.3.3 DISCUSSION OF THE ANTI-TYROSINASE INHIBITION OF THE INVESTIGATED PLANT EXTRACTS

Collection of the nine plant species **1–9** (Table 3.1), which are traditionally used for maintaining the skin structural integrity and healing skin hyperpigmentation was done in Lusikisiki (Ntsubane forest complex) within the E. Cape Province, while plant species **10** (Table 3.1) was collected in Eerste Rivier, which is situated in the W. Cape Province. Initial steps taken prior to the actual execution of the experiments were carried out using the extracts obtained of the suggested plants which included the bio-evaluation of the plant extracts against the tyrosinase enzyme (Figure 3.2 and Figure 3.3).

The results of the skin enzyme inhibitory assay showed that two plant species, namely,  $Rhynchosia\ villosa\$ root and  $Rapanea\$ melanoploes bark inhibited tyrosinase enzyme (Figure 3.2 and Figure 3.3).  $R.\$ villosa\ was found to be the most active extract, as it inhibited mushroom tyrosinase with IC<sub>50</sub> value of 15.67 µg/mL, while  $R.\$ melanoploes inhibited mushroom tyrosinase with IC<sub>50</sub> value of 26.3 µg/mL. Hence  $R.\$ villosa\ was chosen for further chromatographic study, in order to isolate the bioactive compounds.

#### 3.4 DESCRIPTION AND DISTRIBUTION OF R. VILLOSA

#### 3.4.1 DESCRIPTION OF R. VILLOSA

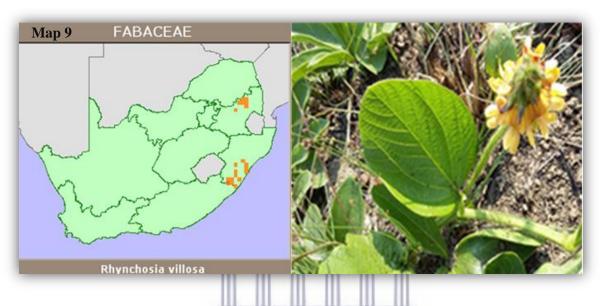
As mentioned previously, *R. villosa* is one of the Fabaceae species. Like many legumes, *R. villosa* is a trailing plant species with hairy petioles, stems and leaves. The flowers of the inflorescence are yellow in colour with red veins, its pinnately compound leaves are light-green and they assume peltate shape. It consists of perennial underground system, an erect rootstock (Figure 3.4), which is the traditional used part for skin depigmentation. Both the rootstock and the leaves of *R. villosa* were bioevaluated in the study against the tyrosinase enzyme, where the rootstock showed higher tyrosinase inhibitory activity than the leaves (Section 3.3); this substantiated its preference in traditional use for skin depigmentation instead of the leaves.



**Figure 3.4**: Illustration of the rootstock of *R. villosa*.

#### 3.4.2 DISTRIBUTION OF R. VILLOSA

As denoted in Table 3.1, *R. villosa* is endemic to S. Africa and widely distributed in KZN, Mpumalanga Province and E. Cape Province (Figure 3.5) (Foden and Potter, 2005).



Map 9: Retrieved from Red List of South Africa Plants website: http://redlist.sanbi.org/species.php?species=370-90

**Figure 3.5**: Distribution of *R. villosa* and its description.

#### 3.5 EXTRACTION AND FRACTIONATION OF R. VILLOSA ROOT

#### 3.5.1 EXTRACTION OF R. VILLOSA ROOT

The *R. villosa* root (RV) with a mass of 131.2 g was initially grated to small shreds before it was allowed to extract in 500 mL of MeOH for 48 hours. The mixture was then filtered and the combined filtrates were concentrated under reduced pressure at 40 °C. The crude extract was then weighed [15.11 g (11.5 % of the raw material)] and kept at approximately 0 °C.

#### 3.5.2 COLUMN CHROMATOGRAPHY OF THE CRUDE EXTRACT

The crude extract of *R. villosa* (15 g) was applied to a silica gel column and eluted with Hex:EtOAc (6.6 L:7.4 L) and EtOAc:MeOH (3.85 L:1.15 L) gradient mixtures of increasing polarity as shown in Table 3.3.

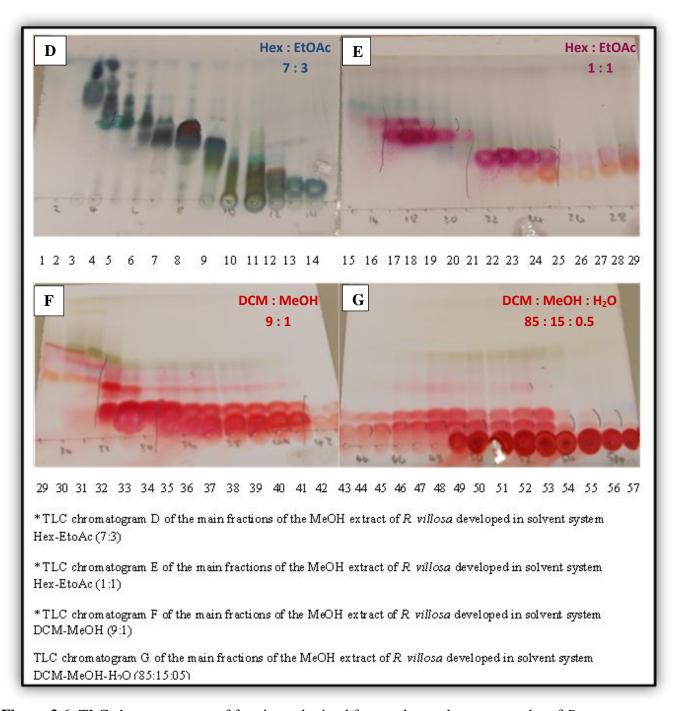
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Table 3.3: Fractionation of the methanolic extract of R. villosa (15 g)

SOLVENT SYSTEM		
Hex:EtOAc	VOLUME (L)	FRACTIONS
100:0	0.5	1-2
90:10	1	3-6
80:20	1	7-10
70:30	2	11-18
60:40	2	19-26
50:50	2	27-34
30:70	2	35-42
10:90	2	43-50
0:100	1.5	51-56
EtOAc:MeOH		
95:5	1	55-58
90:10		59-62
80:20	UNIVERSIT	63-66
70:30	WESTERN	CAP 67-70
50:50	1	71-74

Fractions of 250 mL were collected (Figure 3.6). Similar fractions were combined according to their TLC profile to obtain 20 main fractions (Table 3.4).



**Figure 3.6**: TLC chromatograms of fractions obtained from column chromatography of *R. villosa* methanolic extract,

adsorbent: silica gel

solvent systems: **D** [Hex:EtOAc (7:3)]; **E** [Hex:EtOAc (1:1)]; **F** [DCM:MeOH(9:1)];

**G** [DCM:MeOH:H2O (85:15:0.5)]

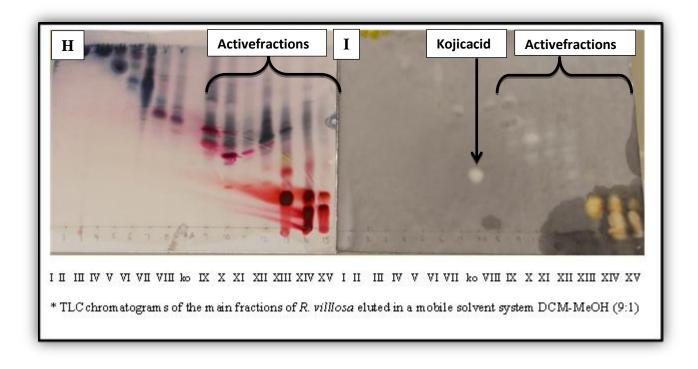
**detection**: sprayed with vanillin/H<sub>2</sub>SO<sub>4</sub> and heated at 110 °C.

Table 3.4: Main fractions of R. villosa methanolic extract

COMBINED	DESIGNATED	
FRACTIONS	CODE	MASS (mg)
1-2	I	5.1
3	II	75.7
4-5	III	79.7
6-7	IV	94.5
8-9	V	46.3
10-12	VI	153.9
13-14	VII	28.8
15-16	VIII	1.2
17-21	IX	118.3
22-25	X	83.0
26-28	XI	10.0
29-31	XII	61.5
32-34	XIIINIVE	RSI 324.2 of the
35-41	XIVEST	ERN 242.0 PE
42-48	XV	163.8
49-53	XVI	612.4
54-59	XVII	4399.4
60-63	XVIII	5295.7
64-67	XIX	583.4
68-70	XX	288.4

# 3.5.3 QUALITATIVE SCREENING OF THE MAIN FRACTIONS USING TLC BIOAUTOGRAPHY

The main fractions (I-XV) obtained from the main column were spotted on TLC plates (2 copies) and developed in DCM:MeOH (9:1). One of the plates (**H**) was sprayed with vanillin/H<sub>2</sub>SO<sub>4</sub> and other plate (**I**) was sprayed with the tyrosinase enzyme and tyrosine, as explained in Section 3.1.7. Figure 3.7 shows the two plates and the inhibition zones in plate **I**, which indicated bioactivity, showing that fractions IX-XV contained the most active compounds.



**Figure 3.7**: TLC chromatograms of the main fractions (I-XV) spotted on TLC plates and developed in DCM:MeOH (9:1), plate **H**: sprayed with vanillin/H<sub>2</sub>SO<sub>4</sub> and heated at 110 °C, plate **I**: sprayed with tyrosinase and L-tyrosine.

### 3.6 ISOLATION OF THE CHEMICAL CONSTITUENTS OF R. VILLOSA

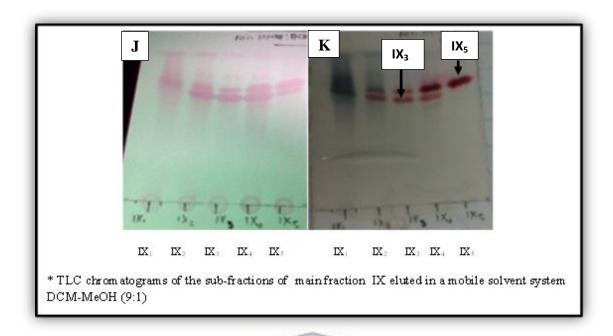
# 3.6.1 COLUMN CHROMATOGRAPHY OF MAIN FRACTION IX

Fraction IX (118.3 mg) was applied onto a sephadex column and eluted with a 5 % aqueous EtOH solution. Similar fractions were combined together to give the sub-fractions  $IX_1 - IX_5$  (Table 3.5; Figure 3.8).

Table 3.5: Collective sub-fractions from the fractionation process of the main extract IX

COMBINED FRACTIONS	DESIGNATED CODE	MASS (mg)
1-12	Discarded	U
13-23	IX <sub>1</sub>	1.1
24-31	IX <sub>2</sub>	15.2
32-34	IX <sub>3</sub>	46.1
36-37	IX <sub>4</sub>	49.8
38-44	IN E IX	6.1

61

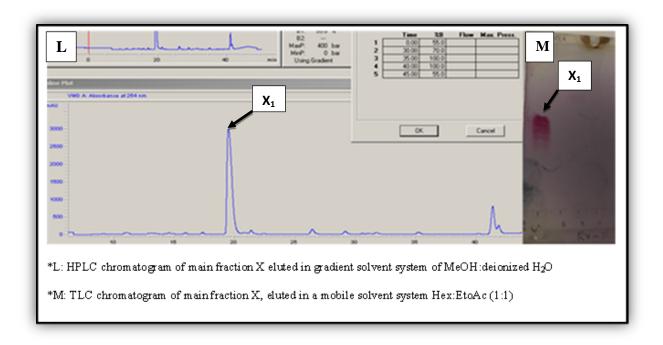


**Figure 3.8**: TLC chromatogram of the sub-fractions of main fraction IX viewed under UV (254 nm; **J**) and after vanillin spraying and heating at 110 °C (**K**).

In order to determine the content and purity of the obtained active compounds and their molecular structures, the samples were subjected to NMR spectroscopy. Thus, it was confirmed from the NMR spectrometer that sub-fraction IX<sub>3</sub> and IX<sub>5</sub> were pure compounds. The samples were isolated with a mass of 46.1 mg (IX<sub>3</sub>; 0.035 %) and 6.1 mg (IX<sub>5</sub>; 0.0046 %), respectively.

#### 3.6.2 HPLC CHROMATOGRAPHY OF MAIN FRACTION X

As indicated in Figure 3.7, the main fraction X was amongst the fractions that showed the highest tyrosinase inhibitory activity. This fraction was initially dissolved in MeOH, before it was injected in the HPLC. A gradient solvent system of MeOH:DIW (55:45 to 70 % in 30 min, then to 100 % in 5 min) was used as the eluent. HPLC of fraction X (83 mg) afforded one prominent peak (Figure 3.9), which was collected and labelled  $X_1$  (4 mg; 0.0030 %).



**Figure 3.9**: HPLC chromatogram (**L**) of the afforded peak  $X_1$ , and TLC chromatogram after spraying with vanillin/ $H_2SO_4$  and heating at 110 °C (**M**).

# \*Conditions used:

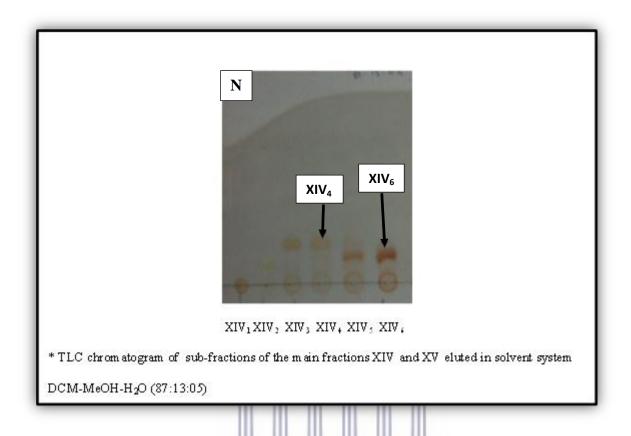
Solvent	MeOH:DIW (55:45 to 70 % in 30 min, then to 100 % in 5 min)
Column	SULPECO, RP-18 ( $25 \times 2.1 \text{ cm}$ )
Flow rate	1.0 mL/min
Detection	UV at 254 nm

#### 3.6.3 COLUMN CHROMATOGRAPHY OF MAIN FRACTIONS (XIV AND XV)

Sephadex fractionation of the two main fractions, viz. XIV and XV with a combined mass of 406 mg was performed. According to the results of the TLC bioautography method which was employed for bioactivity-directed fractionation of the main fractions of *R. villosa* (Figure 3.7), these two polar fractions were found to be amongst active fractions against the tyrosinase enzyme. The combined fraction was applied to a sephadex column using 5 % aqueous EtOH. Two pure compounds labelled XIV<sub>4</sub> (5.9 mg; 0.0045 %) and XIV<sub>6</sub> (287.4 mg; 0.22 %) (Table 3.6; Figure 3.10) were isolated.

**Table 3.6**: Collective sub-fractions from fractionation of the combined main fractions (XIV and  $\overline{XV}$ )

COMBINED FRACTIONS	DESIGNATED CODE	MASS (mg)				
1-30 UN	Discarded	f the				
31-38 W]	ESTEXIVN CA	PE 6.0				
39-42	$XIV_2$	8.5				
43 -54	XIV <sub>3</sub>	68.6				
55-58	$XIV_4$	5.9				
59-61	XIV <sub>5</sub>	20.1				
62-75	$XIV_6$	287.4				



**Figure 3.10**: TLC chromatogram of the sub-fractions of combined main fractions (XIV and XV) after spraying with vanillin/ $H_2SO_4$  and heating at 110°C (N).

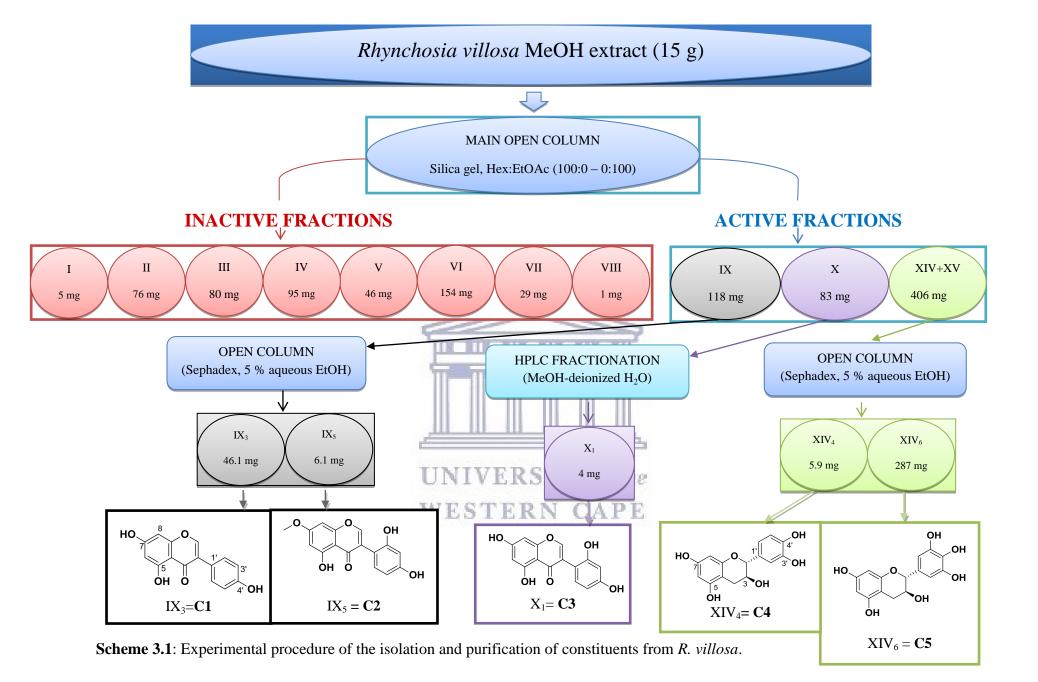
The compounds were thoroughly characterized using NMR spectroscopy. According to the  $^{1}$ H and  $^{13}$ C NMR data obtained, the two polar sub-fractions (XIV<sub>4</sub> and XIV<sub>6</sub>) indicated in Figure 3.10 were found to be pure compounds.

# 3.6.4 SUMMARY OF THE ISOLATION AND PURIFICATION OF CONSTITUENTS FROM R. VILLOSA

As pointed out in Section 3.6.1 – 3.6.3, five compounds viz.  $IX_3$ ,  $IX_5$ ,  $X_1$ ,  $XIV_4$  and  $XIV_6$  were isolated from four active main fractions (IX, X, and combined fractions XIV and XV; Figure 3.7) of R. villosa. To avoid inconvenience, the isolated compounds (IX<sub>3</sub>, IX<sub>5</sub>, X<sub>1</sub>, XIV<sub>4</sub> and XIV<sub>6</sub>) were labelled C1, C2, C3, C4 and C5 from here on respectively. Scheme 3.1 shows the summary of the isolation and purification of constituents from R. villosa. Hence both C1 and C2 were isolated from fractionation of main fraction IX through sephadex open column. Same applies with C4 and C5; they were isolated from fractionation of combined main fractions (XIV and XV) with aid of sephadex open column. While C3 was afforded from fractionation of main

fraction X via semi-prep HPLC.

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#### **CHAPTER FOUR**

# STRUCTURE ELUCIDATION AND ANTI-TYROSINASEACTIVITY OF R. VILLOSA CONSTITUENTS

#### 4.1 STRUCTURE ELUCIDATION OF ISOLATED ISOFLAVONES

## Structure elucidation of genistein (C1):

One-dimensional NMR experiments are fundamental in determining the resonance frequency of each <sup>1</sup>H or <sup>13</sup>C nucleus in the molecule, thus, both <sup>1</sup>H and <sup>13</sup>C NMR experiments were employed the most in elucidating the structures of the isolated compounds in the current study. Moreover, the chemical shifts obtained gave important information about the nature and number of protons and carbons in the compounds of interest. The recorded chemical shifts were then validated by comparing them with existing literature data (Pinheiro and Justino, 2012).

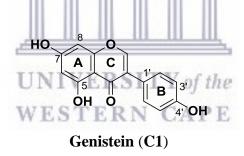


Figure 4.1: Chemical structure of genistein (C1).

Compound IX<sub>3</sub> (46.1 mg; 0.035 %), was isolated as an amorphous yellow powder, showing a characteristic flavonoid nature using UV-Vis light (TLC) and the NMR data obtained. Its spectral data was correlated with the existing literature data (Yoon *et al.*, 2016). The <sup>1</sup>H NMR spectrum (Figure 4.2; Table 4.1) showed aromatic protons of ring A at  $\delta_{\rm H}$  6.02 [(H-6, *d*) 2.0 Hz] and 6.12 [(H-8, *d*) 2.0 Hz], while another deshielded olefinic proton was observed and is attributed to ring C at  $\delta_{\rm H}$  7.86 (H-2, *s*). Signals typically observed for ring B, revealed two pairs

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of equivalent aromatic protons, one pair resides at  $\delta_{\rm H}$  6.72 and 6.74 [(H-3' and H-5', d) 8.8 Hz], while the other pair resides at  $\delta_{\rm H}$  7.23 and 7.25 [(H-2' and H-6', d) 8.8 Hz], which are characteristic of a 1,4-disubstituted ring B.

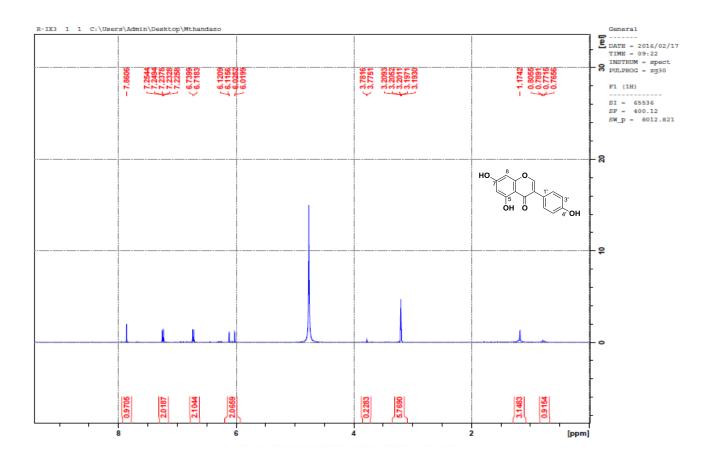
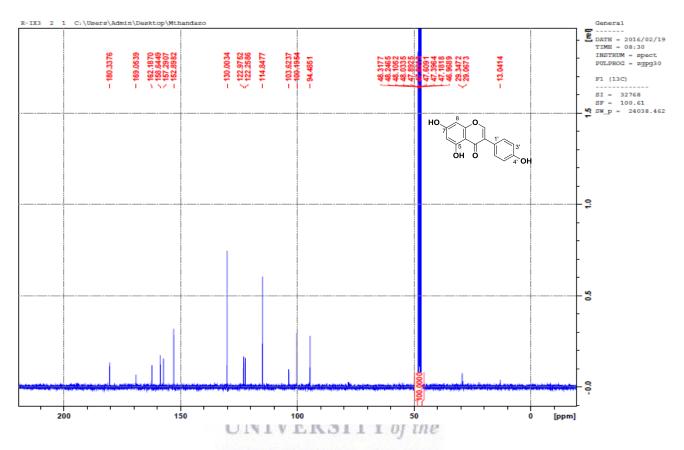


Figure 4.2: <sup>1</sup>H NMR spectrum of compound C1 in CD<sub>3</sub>OD.

The  $^{13}$ C NMR spectrum (Figure 4.3; Table 4.1) depicted twelve aromatic carbons  $\delta_c$  162.2 (C-5), 100.2 (C-6), 169.1 (C-7), 94.8 (C-8), 157.3 (C-9), 103.6 (C-10), 122.3 (C-1'), 158.6 (C-4'), 130.0 (C-2' and C-6'), 114.8 (C-3' and C-5') belonging to ring A and B. An olefinic carbon at  $\delta_c$  152.9 (C-2), quaternary carbon at  $\delta_c$ 123.0 (C-3) and carbonyl carbon at  $\delta_c$  180.3 (C-4, C=O) were observed and are attributed to ring C. The high intensity signals at  $\delta_C$  114.8 and 130.0 were

evidence of the presence of the two pairs of equivalent carbons, i.e., C-3' with C-5' and C-2' with C-6' positions, respectively.



**Figure 4.3**:<sup>13</sup>C NMR spectrum of compound **C1** in CD<sub>3</sub>OD.

These data implied the presence of an isoflavone nucleus functionalized with a 5,7,4'-trihydroxy group. Careful review of the literature thus confirmed the compound to be genistein, a common isoflavone isolated from soybean and widely distributed in Leguminosae. Figure 4.1 shows the chemical structure of genistein (C1).

# Structure elucidation of cajanin (C2):

Elucidation of the chemical structure of **C2** was achieved by employing heteronuclear 2D NMR experiments (HSQC and HMBC), which generate contour maps that show the correlations between proton and carbon nuclei, hence providing an opportunity to observe the <sup>1</sup>H and <sup>13</sup>C NMR data directly (Claridge, 1999). Two-dimensional NOESY data were also acquired and these data are useful at establishing non-bonded connectivity, or connectivity through space (Pinheiro and Justino, 2012).

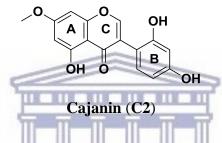


Figure 4.4: Chemical structure of cajanin (C2).

Compound IX<sub>5</sub> (6.1 mg; 0.0046 %) was isolated as an amorphous yellow powder. The  $^{1}$ H NMR spectrum (Figure 4.5; Table 4.1) showed two aromatic protons attributed to ring A  $\delta_{\rm H}$  6.26 [(H-6, d) 2.2 Hz], 6.46 [(H-8, d) 2.2 Hz] and three methyl protons  $\delta_{\rm H}$  3.81 (MeO-7, s) attributed to a methoxy functional group. An additional three protons were observed for ring B at  $\delta_{\rm H}$  6.30 [(H-3', d) 2.4 Hz], 6.28 [(H-5', dd) 2.4 Hz, 8.4 Hz], and 6.96 [(H-6', d) 8.4 Hz], while one deshielded proton was observed to resonate at  $\delta_{\rm H}$  7.91 (H-2, s) and is attributed to ring C.

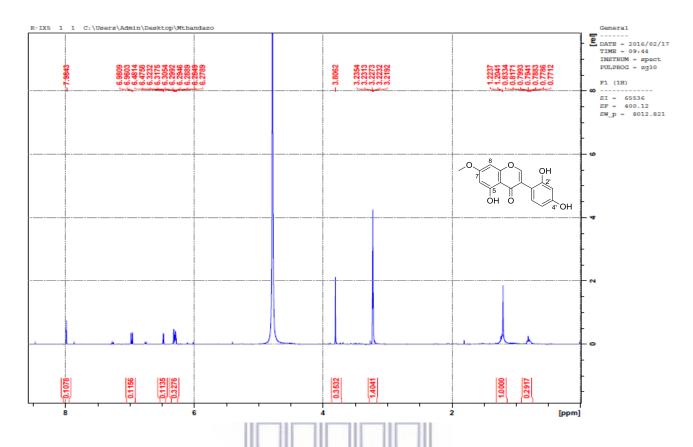
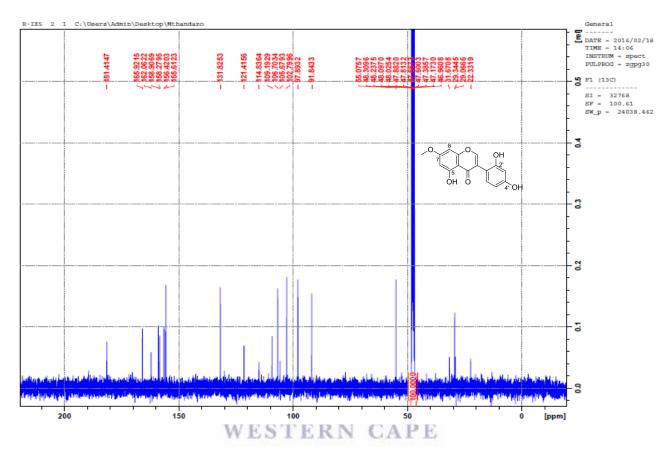


Figure 4.5: <sup>1</sup>H NMR spectrum of compound C2 in CD<sub>3</sub>OD.

Significant differences were observed in the  $^{13}$ C NMR spectrum (Figure 4.6; Table 4.1) of C2 when compared to that of C1 as some of the ring B carbons were found to be shielded and this is likely due to additional electron density imparted by the hydroxyl group occurred at C-2' position. Two carbon signals residing at  $\delta_c$  102.8 (C-3'; *ortho*) and 106.6 (C-5'; *para*) were highly shielded in comparison to C1, while both C-6' ( $\delta_c$  131.8; *meta*) and C-4' ( $\delta_c$  158.6; *meta*) were found in the same positions as for C1. As mentioned above, unlike in C1, a deshielded carbon signal occurring at  $\delta_c$  156.2 (C-2') indicated the presence of a hydroxyl functional group at the C-2' position. Additional carbon signals were observed to resonate at  $\delta_c$  155.6 (C-2), 121, 4 (C-3), 181.4 (C-4), 162.1 (C-5), 97.9 (C-6), 165.9 (C-7), 91.8 (C-8), 158.3 (C-9), 105.6 (C-10)

109.2 (C-1') and 55.1 (MeO-7) accounting for all the expected signals for the structure attributed to C2 (Figure 4.4).



**Figure 4.6**:<sup>13</sup>C NMR spectrum of compound **C2** in CD<sub>3</sub>OD.

The NMR spectra of **C2** was found to be similar to that of **C1**, with the main difference being the presence of the methoxy signals at  $\delta_H$  3.81/ $\delta_c$  55.1 in  $^1H$  and  $^{13}C$  NMR spectra, respectively, indicating the presence of the methoxy group (OMe) at C-7. The position of the methoxy group was additionally evidenced from a 2D HMBC spectrum (Figure 4.7) which showed a correlation between the *O*-methyl protons at  $\delta_H$  3.81, to C-7 ( $\delta_C$  165.9).

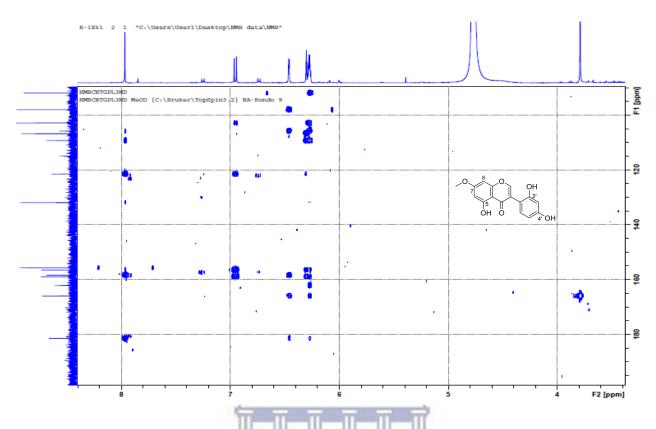


Figure 4.7: HMBC NMR spectrum of compound C2 in CD<sub>3</sub>OD.

Through <sup>13</sup>C NMR (Figure 4.6) and 2D HSQC spectra (Figure 4.8), we were able to assign unambiguously the proton/carbon correlations and other unprotonated signals.

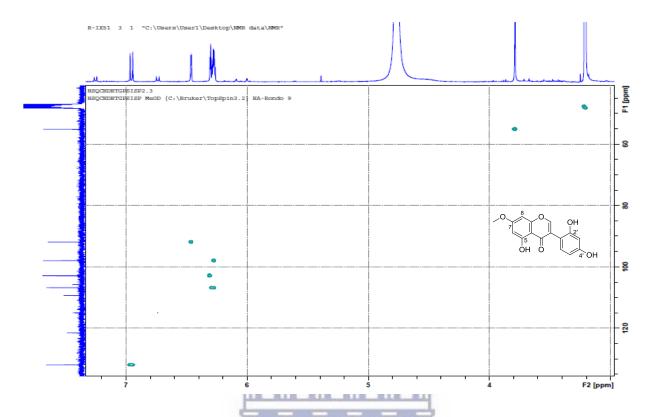


Figure 4.8: HSQC NMR spectrum of compound C2 in CD<sub>3</sub>OD.

In addition, the methoxy group in the 2D NOESY spectrum (Figure 4.9) showed cross peaks with H-6 and H-8. The chemical structure of **C2** was finally confirmed by comparison with the previously reported data (Waffo *et al.*, 2000). Compound **C2** was identified as cajanin, a major constituent of one of the closely related species to that of *Rhynchosia*, *Cajanus cajan*.

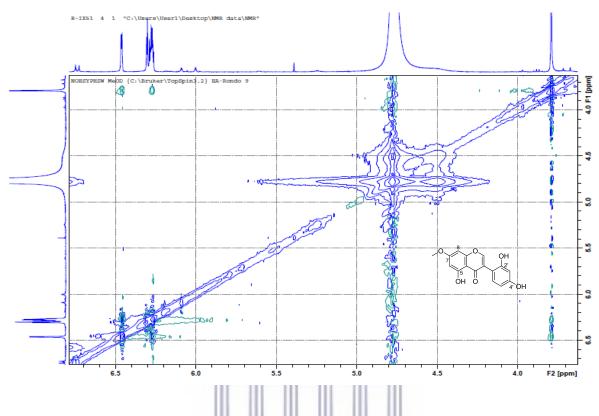


Figure 4.9: NOESY NMR spectrum of compound C2 in CD<sub>3</sub>OD.

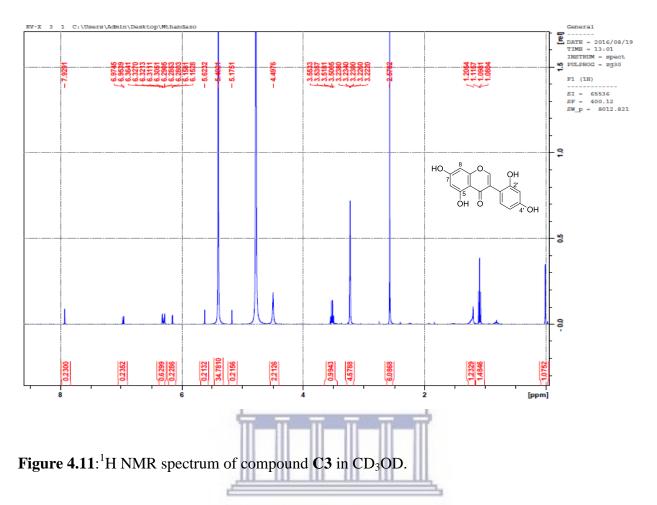
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# Structure elucidation of 2'-hydroxygenistein (C3):

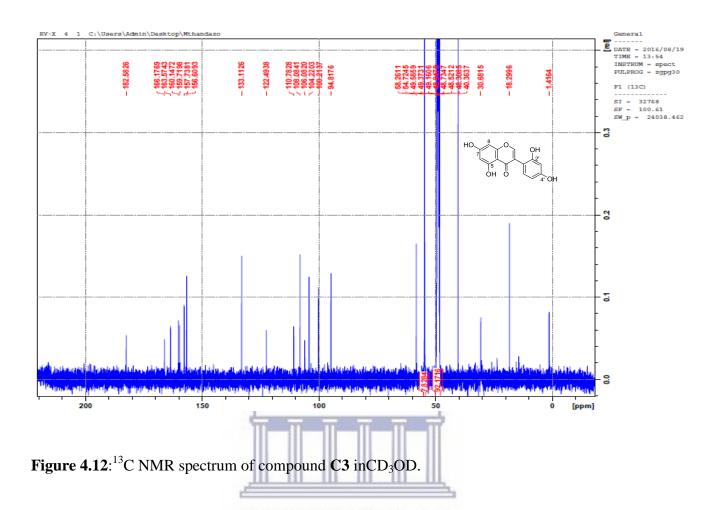
2'-hydroxygenistein (C3)

**Figure 4.10**: Chemical structure of 2'-hydroxygenistein (C3).

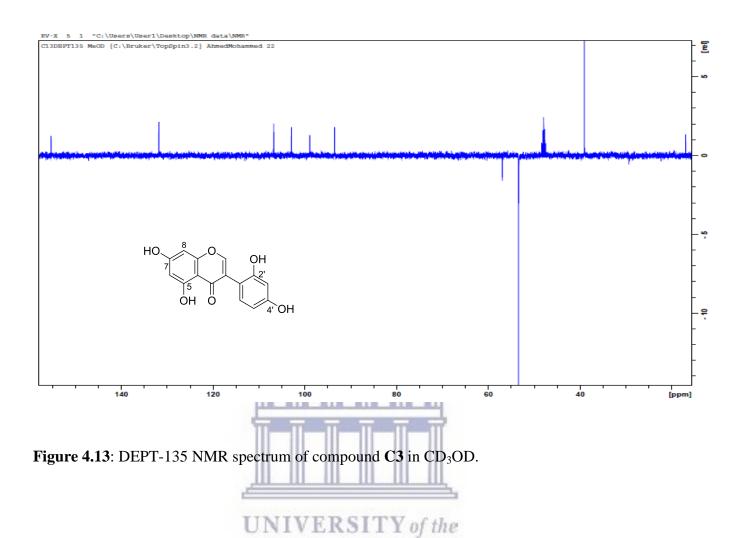
Compound  $X_1$  (4mg; 0.0030 %) was obtained as a pale-yellow powder, and its spectral data was found to be very similar to that of C2, except the methoxy signals were absent. The structure elucidation of C3 (Figure 4.10) was carried out by comparing its spectral data with the existing, documented data (Awouafack *et al.*, 2011; Lo *et al.*, 2002). The <sup>1</sup>H-NMR spectrum (Figure 4.11; Table 4.1) showed the two deshielded protons residing at  $\delta_H$  6. 14 [(H-6, d) 2.0 Hz], and 6.27 [(H-8, d) 2.0 Hz] to be a *meta* coupled pair that revealed free hydroxyl groups at the C-5 and C-7 positions. While the protons in ring B were observed at  $\delta_H$  6.31 [(H-3', d) 2.4 Hz], 6.28 [(H-5', dd) 2.0 Hz, 8.4 Hz], and 6.97 [(H-6', d) 8.4 Hz]. A deshielded singlet that is characteristic of all isoflavones, signifying the olefinic proton in ring C, was observed at  $\delta_H$  7.93 (H-2, s).



The  $^{13}$ C NMR spectrum (Figure 4.12; Table 4.1) showed twelve aromatic carbons  $\delta_c$  163.6 (C-5), 100.2 (C-6), 162.2 (C-7), 94.8 (C-8), 159.7 (C-9), 106.1 (C-10), 110.8 (C-1'), 160.2 (C-2'), 114.8 (C-3'), 157.7 (C-4'), 108.1 (C-5') and 133.1 (C-6') of ring A and B. An olefinic carbon residing at  $\delta_c$  156.6 (C-2), quaternary carbon at  $\delta_c$  122.5 (C-3) and carbonyl carbon at  $\delta_c$  182.5 (C-4, C=O) were observed and attributed to ring C.



Moreover, DEPT-135 spectrum (Figure 4.13) aided in assigning protonated carbons and proton/carbon correlations. From the spectral data of **C3** and through to careful literature review, the compound was thus identified as 2'-hydroxygenistein (Figure 4.10).



WESTERN CAPE

#### 4.2 STRUCTURE ELUCIDATION OF THE ISOLATED FLAVAN-3-OLS

# Structure elucidation of catechin (C4):

#### Catechin (C4)

Figure 4.14: Chemical structure of catechin (C4).

Compound XIV<sub>4</sub> (5.9 mg; 0.0045 %) was obtained as a dark yellow amorphous powder, and it was identified by comparing the obtained spectral data with the previously reported data (Hye *et al.*, 2009). The <sup>1</sup>H-NMR spectrum (Figure 4.15; Table 4.1) showed peaks at  $\delta_{\rm H}$  4.51 [H-2, d, J(H-2 $\alpha$ , H-3 $\alpha$ ) 6.1 Hz]; 3.91 (H-3, m); 2.79 [H-4 $\alpha$ , dd, J(H-4 $\alpha$ , H-3 $\alpha$ ) 8 Hz, J(H-4 $\alpha$ , H-4 $\beta$ ) 16 Hz]; 2.44 [H-4 $\beta$ , dd, J(H-4 $\beta$ , H-3 $\alpha$ ) 5.6 Hz, J(H-4 $\beta$ , H-4 $\alpha$ ) 16 Hz]; 5.78 [H-6, d, J(H-6, H-8) 2.4 Hz]; 5.84 [H-8, d, J(H-8, H-6) 2.4 Hz]; 6.75 [H-2', d, J(H-2', H-6') 1.8 Hz]; 6.69 [H-5', d, J(H-5', H-6') 8 Hz] and 6.64 [H-6', dd, J(H-6', H-2') 1.8 Hz, J(H-6', H-5') 8 Hz]. A strong coupling constant (J) between  $\delta_{\rm H}$  4.51 (H-2 $\alpha$ ) and 3.91 (H-3 $\alpha$ ) of 6.1 Hz was observed, supporting a *trans*-configuration between these two protons (Foo, 1987).

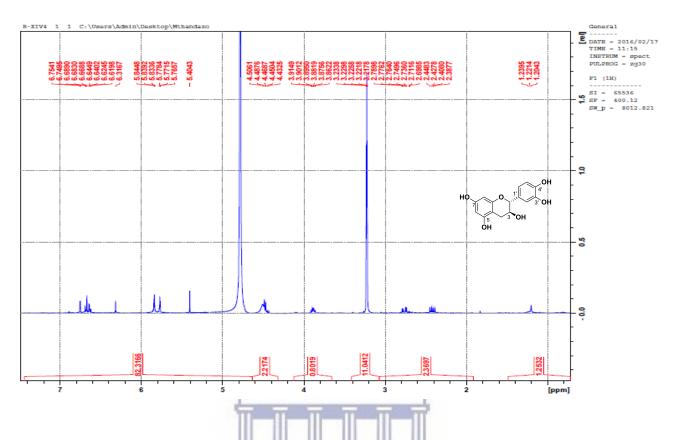
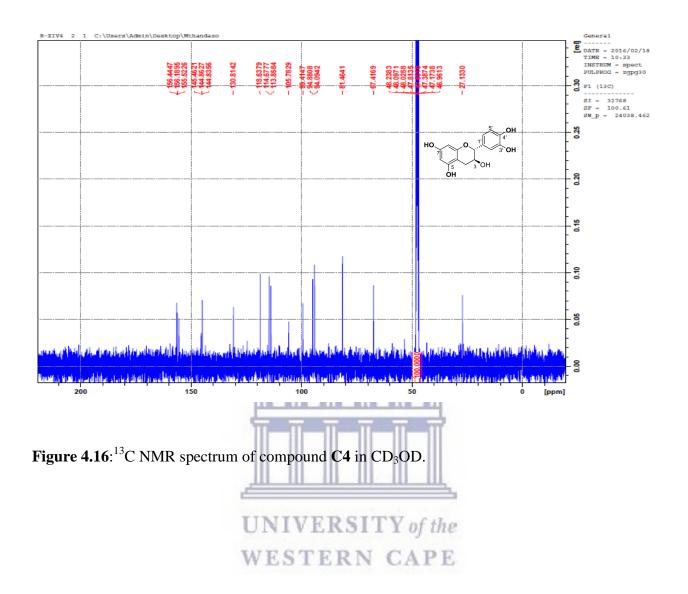


Figure 4.15: <sup>1</sup>H NMR spectrum of compound C4 in CD<sub>3</sub>OD.

The  $^{13}$ C NMR spectrum (Figure 4.16; Table 4.1) of **C4** showed carbon signals at  $\delta_{C}$  27.1 (C-4), 67.4 (C-3), 81.5 (C-2), 94.1 (C-6), 94.9 (C-8), 99.4 (C-10), 113.9 (C-2'), 114.7 (C-5'), 118.6 (C-6'), 130.8 (C-1'), 144.8 (C-3'), 144.9 (C-4'), 155.5 (C-5), 156.2 (C-9) and 156.4 (C-7). The above data showed that **C4** consists of a flavan-3-ol backbone, through a thorough literature review; it was identified as catechin, a common procyanidin monomer typically isolated from green tea (Jeong and Kong, 2004).



# Structure elucidation of gallocatechin (C5):

## Gallocatechin (C5)

**Figure 4.17**: Chemical structure of gallocatechin(**C5**).

Conversely, compound XIV<sub>6</sub> (287.4 mg; 0.22 %) was obtained as a brownish-red powder, and it demonstrated very similar NMR spectra to that of **C4**, with only slight differences. The <sup>1</sup>H-NMR spectrum (Figure 4.18; Table 4.1) showed proton signals at  $\delta_{\rm H}$  4.48 [H-2, d, J(H-2 $\alpha$ , H-3 $\beta$ ) 7.2 Hz]; 3.93 (H-3, m); 2.46 [H-4 $\alpha$ , dd, J(H-4 $\alpha$ , H-3 $\alpha$ ) 7.8 Hz, J(H-4 $\alpha$ , H-4 $\beta$ ) 16.2 Hz]; 2.75 [H-4 $\beta$ , dd, J(H-4 $\beta$ , H-3 $\alpha$ ) 5.2 Hz, J(H-4 $\beta$ , H-4 $\beta$ ) 16.2 Hz]; 5.86 [H-6, d, J(H-6, H-8) 2.0 Hz]; 5.84 [H-8, d, J(H-8, H-6) 2.0 Hz] and 6.33 (H-2' and H-6', s). The intense signal at  $\delta_{\rm H}$  6.33 signifies the presence of two equivalent aromatic protons at the C-2' and C-6' positions in ring B. This peak was found to be absent in the case of **C4**. Likewise as in the case of **C4**, a strong coupling constant of 7.2 Hz was observed between  $\delta_{\rm H}$  4.48 (H-2 $\alpha$ ) and 3.93 (H-3 $\alpha$ ), signifying a *trans*-configuration between the two protons. In the case of both **C4** and **C5** epimers, the coupling constant between these signals would be very weak (Foo, 1987).

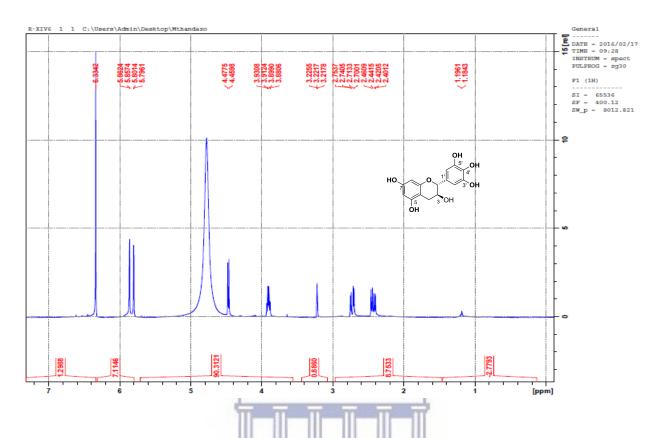


Figure 4.18: <sup>1</sup>H NMR spectrum of compound C5 in CD<sub>3</sub>OD.

The  $^{13}$ C NMR spectrum (Figure 4.19; Table 4.1) revealed carbon signals at  $\delta_{C}$  26.6 (C-4), 67.3 (C-3), 81.4 (C-2), 94.2 (C-6), 95.0 (C-8), 99.4 (C-10), 105.9 (C-2' and C-6'), 130.2 (C-1'), 132.6 (C-4'), 145.4 (C-3' and C-5'), 155.4 (C-5), 156.2 (C-9) and 156.3 (C-7). The high intensity signals at  $\delta_{C}$  105.9 and 145.4, reveals the presence of two pairs of equivalent carbons i.e., C-2' with C-6', and C-3' with C-5', respectively. Another distinction for the spectra of **C5**, was a carbon signal residing at  $\delta_{C}$  132.6 signifying that C-4' was highly shielded in comparison to the **C4** spectra, where the C-4' signal was found at  $\delta_{C}$  144.9. This is likely due to the additional electron density provided by the hydroxyl group at the C-5' position.

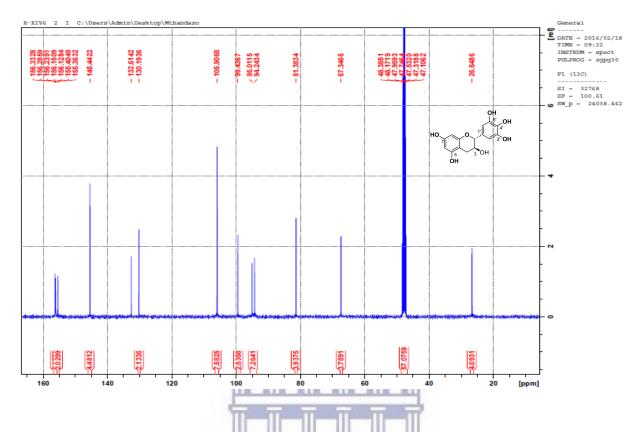


Figure 4.19:<sup>13</sup>C NMR spectrum of compound C5 in CD<sub>3</sub>OD.

Moreover, the unusual dark brown colour of this pure compound (C5) is thought to be due to the presence of pyrogallol moiety (ring B) (Kumada and Kato, 1970). Such compounds are typically a white solid, but its samples and derivative samples (e.g., C5) are typically brownish because of its sensitivity toward oxygen (Fiege *et al.*, 2014). The chemical structure of C5 was finally confirmed by comparison with the previously reported data (Medimagh *et al.*, 2010) and identified as gallocatechin (Figure 4.17).

**Table 4.1**:  ${}^{1}$ H [400 MHz: m, J(Hz)] and  ${}^{13}$ C (100 MHz) NMR spectral data of isolated compounds (C1 - C5) in CD<sub>3</sub>OD

	I				ı				1		
Code &											
No.	]	$X_3$ (C1)	$IX_5$ (C2)			$X_1$ (C3)		XIV <sub>4</sub> ( <b>C4</b> )		XIV <sub>6</sub> (C5)	
	но 8	0_			но, ஃ	HO, & O,		5'		он	
	7	1'~	.0. &	.O.	OH 2'		HO. Q. 11 4'		5' OH		
	\[ \] 5   OI	3'	7	OH 2'	\ref{5}   OI	H 0	7	3, OH	HO 7	0 3 ОН	
Chemical		4 OH	) 5 <b>O</b> H			4 <sup>-</sup> OH	OI	3 OH	5	3 ОН	
structure				4 <sup>-</sup> OH				•	ОН		
	<sup>13</sup> C	$^{1}\mathbf{H}, m, J$	<sup>13</sup> C	$^{1}\mathbf{H}, m, J$	<sup>13</sup> C	$^{1}\mathbf{H}, m, J$	<sup>13</sup> C	$^{1}\mathbf{H}, m, J$	<sup>13</sup> C	$^{1}\mathbf{H}, m, J$	
2	152.9	<b>7.86</b> , <i>s</i>	155.6	<b>7.91</b> , <i>s</i>	156.6	<b>7.93</b> , s	81.5	<b>4.51</b> , d, 6.1	81.4	<b>4.48</b> , <i>d</i> , 7.2	
3	123.0		121.4		122.5		67.4	<b>3.91</b> , <i>m</i>	67.3	<b>3.93</b> , <i>m</i>	
								H-4α: 2.44, dd,		H-4α: 2.46, dd,	
								8.0, 16.0 <b>H-4β</b> : <b>2.79</b> , <i>dd</i> ,		7.8, 16.2 <b>H-4β</b> : <b>2.75</b> , <i>dd</i> ,	
4	180.3		181.4		182.5		27.1	5.6, 16.0	26.6	5.2, 16.2	
5	162.2		162.1		163.6		155.5		155.4		
6	100.2	<b>6.02</b> , d, 2.0	97.9	<b>6.26</b> , d, 22	100.2	<b>6.14</b> , d, 2.0	94.1	<b>5.77</b> , <i>d</i> , 2.4	94.2	<b>5.80</b> , <i>d</i> , 2.0	
7	169.1		165.9	311-11	166.2	11-11-11	156.4		156.3		
8	94.8	<b>6.12</b> , d, 2.0	91.8	<b>6.46</b> , d, 2.2	94.8	<b>6.27</b> , d, 2.0	94.9	<b>5.84</b> , <i>d</i> , 2.4	95.0	<b>5.86</b> , <i>d</i> , 2.0	
9	157.3		158.3		159.7		156.2		156.2		
10	103.6		105.6	,111	106.1	111 111 11	99.4		99.4		
1'	122.3		109.2	******	110.8	THE PART OF	130.8		130.2		
2'	130.0	<b>7.23</b> , d, 8.8	156.4	UNIT	157.7	SIIY of ti	113.9	<b>6.75</b> , <i>d</i> , 1.8	105.9	<b>6.33</b> , s	
3'	114.8	<b>6.72</b> , d, 8.8	102.8	<b>6.30</b> , <i>d</i> , 2.4	104.2	<b>6.31</b> , d, 2.4	144.8		145.4		
4'	158.6		158.9	20	160.2		144.9		132.6		
	1110	( <b>54</b> 100	104	<b>6.28</b> , <i>dd</i> , 2.4,	100.1	<b>6.28</b> , <i>dd</i> , 2.4,	444=	((0, 1, 0, 0	147.4		
5'	114.8	<b>6.74</b> , d, 8.8	106.5	8.4	108.1	8.4	114.7	<b>6.69</b> , d, 8.0 <b>6.64</b> , dd, 1.8,	145.4		
6'	130.0	<b>7.25</b> , <i>d</i> , 8.8	131.8	<b>6.96</b> , <i>d</i> , 8.4	133.1	<b>6.97</b> , d, 8.4	118.6	8.0	105.9	<b>6.33</b> , s	
OMe			55.1	<b>3.81</b> , <i>s</i>							

#### 4.3 ANTI-TYROSINASE ACTIVITIES OF THE ISOLATED COMPOUNDS

Tyrosinase inhibitors play an important role in healing and preventing pigmentation disorders in relation to hyperpigmentation, as they inhibit the overproduction of the tyrosinase enzyme, consequently leading to the adequate regulation of the production of melanin by the melanocytes. The effect of browning of food, due to the tyrosinase enzymatic effects has already been discussed, as within the broader aspects of the present study. Thus, the tyrosinase inhibitors are also required to inhibit and prevent the adverse effects of browning during food storage, processing and consumption, so as to maintain the commercial value or quality of such products. It has been shown that the free 1,3-dihydroxy system (ring A) in the isoflavone structure strongly affects both the inhibitory strength and the inhibitory mode of the isoflavones to mushroom tyrosinase (Chang, 2014). Therefore, the present study warrants the study of the anti-tyrosinase activity, as all the isolated compounds in this study consisted of a free 1,3-dihydroxy system and were all indeed found to be active against the tyrosinase enzyme. The IC<sub>50</sub> values ranged between 31.5 and 69.5  $\mu$ M (Figure 4.20; Table 4.2), showing the anti-tyrosinase effectiveness of the isolated compounds to be in the following order; C1>C4>C5>C5> C3.

It was extrapolated that substitution of hydroxy group by a methoxy group has a significant effect on the tyrosinase activity as shown by cajanin (C2), which is an O-methylated derivative of 2'-hydroxygenistein (C3), as it revealed a higher activity than its hydroxylated analogue. Therefore, our study correlates the existing data, which has proven that methoxylation play a vital role in inhibition strength of the certain methoxylated compound to the mushroom tyrosinase activity (Kim *et al.*, 2002). The tyrosinase inhibitory assay was performed as described in Section 3.1.6. Tyrosinase inhibitory effects of the isolated constituents from R. *villosa* are shown in Table 4.2 and portrayed in Figure 4.20.

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Table 4.2: Tyrosinase inhibition of R. villosa constituents

		Average percentages of tyrosinase inhibition (%) in various concentrations ( $\mu g/mL$ ) and anti-tyrosinase IC <sub>50</sub> values ( $\mu g/mL \& \mu M$ ) of the isolated compounds						
Compound code & No.	Name of the isolated compound	100.0 μg/mL	50.0 μg/mL	25.0 μg/mL	12.5 μg/mL	6.25 μg/mL	IC <sub>50</sub> (μg/mL)	IC <sub>50</sub> (μ <b>M</b> )
IX <sub>3</sub> (C1)	Genistein	69.47	62.56	59.40	42.99	36.52	8.5	31.45
IX <sub>5</sub> ( <b>C2</b> )	Cajanin	66.27	59.46	50.31	42.12	36.95	11.7	38.97
X <sub>1</sub> (C3)	2'-hydroxygenistein	64.38	47.80	42.62	32.09	19.67	19.9	69.49
XIV <sub>4</sub> ( <b>C4</b> )	Catechin	75.83	68.10	51.81	41.24	28.77	10.7	36.86
XIV <sub>6</sub> ( <b>C5</b> )	Gallocatechin	72.46	52.30	37.52	21.03	19.09	18.5	60.40
Ko	Kojic acid	98.48	97.74	94.15	89.32	87.02	0.9	6.33

\*Ko: Kojic acid was used as a positive control

# Anti-tyrosinase activity ( $\mu M$ ) of $\it R.~\it villosa$ constituents

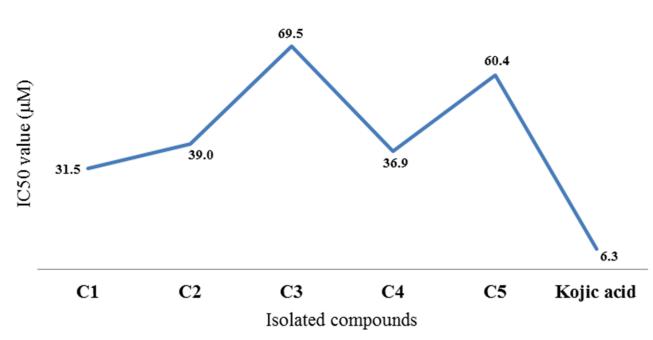


Figure 4.20: Effects of R. villosa constituents on inhibition of tyrosinase activity.

# 4.4 DISCUSSION ON THE ANTI-TYROSINASE ACTIVITIES OF THE ISOLATED COMPOUNDS

The tyrosinase enzyme assay and its qualitative method (TLC bioautography), were used for the bioactivity-directed fractionation and isolation of the five active constituents in the present study. Through TLC bioautography (Figure 3.7) it was observed that main fractions IX - XV were the most active main fractions as they depicted clear white zones of tyrosinase inhibition. The isolated constituents were also bio-evaluated against the tyrosinase enzyme, where compound C1 showed the highest tyrosinase inhibitory activity out of the isolated compounds. Generally, all the isolated compounds showed high activity against the mushroom tyrosinase, with  $IC_{50}$  value differences fluctuating from 4 - 8.5 fold less than that of kojic acid (positive control).

Furthermore, the Fabaceae species are widely used in pharmaceutical and/or cosmetics, one of such species, which consists of the beneficial effects on skin, through its anti-tyrosinase, antioxidant and anti-inflammatory activities is that of *Glycine max*, well-known as soybean. It has been found that the effects of soybean are due to existence of isoflavones within it (Shamani, 2008). From the results of the present study, a similarity is distinguished between the soybean and *R. villosa*, since *R. villosa* also showed a large number of its constituents to be isoflavones; hence this supports its use as a traditional cosmeceutical.

Genistein (C1) (Perkin and Newbury, 1899) is said to be one of the most common isoflovones found in skin care products that are made from soybeans. Such skin care products brighten the skin, decrease redness, boost collagen production and improve skin tone (Jaliman, 2011). Unlike many synthetic substances that are applied on the skin, topically applied soybean does not exert any hormonal effects (Schmid *et al.*, 2008). Our findings confirm the existing data, which demonstrated that genistein (C1) is a natural tyrosinase inhibitor, as it was found to inhibit

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mushroom tyrosinase activity with an IC<sub>50</sub> value of 31.5  $\mu$ M, only 4-fold less than that of kojic acid, an accepted standard tyrosinase inhibitor (6.3  $\mu$ M).

Scientific findings attest to the fact that isoflavones consists of estrogenic hormonal activity, since they have a similar structure with that of the human female hormone 17- $\beta$ -estradiol. They can bind to both  $\alpha$  and  $\beta$  estrogen receptors and they can imitate the action of the endogenous estrogens on target organs, thereby exerting many health benefits when used in some hormone-dependent diseases (Vitale *et al.*, 2013). Hence, post-menopausal women, whose estrogen production has dropped dramatically, may see obvious results when using soybean-containing moisturizing lotions, since the soybean natural phytoestrogens work to repair the thinness of the skin and the decreased collagen production caused by the lack of estrogen (Schmid *et al.*, 2001). This confirms the use of soybean and *R. villosa* as a form of natural relief of estrogenic symptoms and/or other hormone-dependent diseases. Also, the effectiveness of soybean isoflavones in healing dermatological and hormone-dependant disorders validates the fact that the two plant species (*Glycine max* and *R. villosa*) can be used interchangeably, as they have similar biological and pharmacological activities.

Genistein (C1) consists of derivatives such as the well-known *ortho*-hydroxygenisteins (OHGs) (Chang, 2014; Chang, 2009). These derivatives are typically isolated from fermented soybean foods, microbial fermentation broth feeding with soybean meal or produced by genetically modified microorganisms and they are rarely found in plant species (Chang, 2014). The *ortho*-hydroxylated derivatives of genistein (C1) differ in structure-activity relationships, thus, the number and positions of hydroxyl groups in the chemical structures of such isoflavones intensely affect their function (Scotti *et al.*, 2012). For instance, in the current experiment, 2'-hydroxygenistein (C3) (Ingham, 1977), which is one of the OHGs, showed a lower activity when

compared to other isolated constituents, with an IC<sub>50</sub> value of 69.5  $\mu$ M, although it comprises of a free 1,3-dihydroxy system (ring A), together with a resorcinol moiety (ring B). Its analogue, 3'-hydroxygenistein, however, has been reported to consist of an anti-tyrosinase IC<sub>50</sub> value of 15.9  $\mu$ M, merely 1.5 fold less than that of the control, kojic acid (Ding *et al.*, 2015).

Cajanin (C2) (Bilyard et al., 1984) is the major constituent of Cajanus cajan, a species from Cajanus genus; Cajanus is one of the Cajaninae genera and it is very closely related to Rhynchosia and other two genera from this subtribe, Atylosia and Dunbaria (Pundir and Singh, 1985; Pundir and Singh, 1986). The species Cajanus cajan has been found to have potent antioxidant, antiplasmodial, anti-inflammatory and hypocholesterolemic activities (Rerk-am et al., 2013, Pal et al., 2011). Rerk-am et al. (2013) revealed that although C. cajan seeds had a slight anti-tyrosinase activity with EC<sub>50</sub> value of 267 µg/mL against the mushroom tyrosinase, it quenched DPPH radicals with EC<sub>50</sub> value of 44.36 µg/mL, while it demonstrated substantial activity against the inhibition of the Fe<sup>2+</sup>-induced microsomal lipid peroxidation with EC<sub>50</sub> of 25.97 µg/mL. The latter data substantiates the high potential of C. cajan constituents and it also proves that such constituents can be beneficial in healing hyperpigmentation related disorders, particularly acne vulgaris due to the plant species' high inhibition of the Fe<sup>2+</sup>-induced microsomal lipid peroxidation (Bowe and Logan, 2010). The current study validated this postulation as the cajanin (C2) isolated from R. villosa was revealed to inhibit the mushroom tyrosinase with a substantial IC $_{50}$  value of 39.0  $\mu M$ .

On the other hand, the present study totally disagrees with that of Meyer (1960), which revealed that catechin (C4) (Roux and Maihs, 1958) has a tanning property with respect to the human skin, and suggested that it may be a useful active ingredient for the treatment of hypopigmentation, leukoderma. Our study revealed catechin (C4) to have a high anti-

melonogenesis effect with an IC<sub>50</sub> value of 36.9  $\mu$ M against the mushroom tyrosinase, proving that it may perform adequately as a depigmenting agent. Our results are substantiated by that of Sato and Toriyama (2009), as they found that catechin (**C4**) inhibited melanin synthesis in B16 melanoma cells. Furthermore, Western blotting analysis was performed in their study which further indicated characteristics of the catechin (**C4**) inhibition of tyrosinase.

The current study is consistent with the previous data (Ko *et al.*, 2011), as they found out that gallocatechin (C5) (Saayman and Roux, 1965) has a tyrosinase inhibition percentage of 66.8 %, when screened at 100 μg/mL, and 55.7 % at 50 μg/mL. Therefore, in our study gallocatechin (C5) had a comparable tyrosinase inhibition percentage of 72.5 % when screened at 100 μg/mL, and 52.3 % at 50 μg/mL. Its epimer, epigallocatechin was reported to consist of an anti-tyrosinase IC<sub>50</sub> value of 0.7 mM by Ondonbayar *et al.* (2016), 11-fold less active when compared to that of gallocatechin (C5), which was found to be 60.4 μM in the current study. It can be postulated that the difference between catechin (C4; 36.9 μM) and gallocatechin (C5; 60.4 μM) anti-tyrosinase activities of nearly 23.5 μM (0.55-fold) is due to the presence of pyrogallol moiety (ring B) in gallocatechin (C5).

Generally, the above-mentioned types of flavonoids are frequently referred to as green tea polyphenols (GTPs) due to their abundance in green tea made from *Camellia sinensis* leaves (Rahmani *et al.*, 2015). Research has revealed that the modes of action of GTPs for the photo protection of the skin include ROS scavenging and regulation of genes to inhibit apoptosis, reduce inflammation, improve the skin barrier, and to repair DNA (Hsu, 2005). Our results are consistent with previous documented data, since both compound **C4** and **C5** were found to be active against the mushroom tyrosinase enzyme. Additionally, all the isolated compounds were

active against mushroom tyrosinase. Hence this supports the traditional uses of *R. villosa* for skin depigmentation.



### **CHAPTER FIVE**

## CONCLUSION AND RECOMMENDATIONS

#### 5.1 CONCLUSION

Preliminary screening of the ten investigated plant species, viz. *Cassine peragua*, *Cassipourea gummiflua*, *Clivia miniata*, *Cryptocarya myrtifolia*, *Gunnera perpensa*, *Kigelia africana*, *Protorhus longifolia*, *Rapanea melanophloeos*, *Rhynchosia villosa*, and *Senecio serratuloides* against the tyrosinase enzyme showed that *R. villosa* was the most active plant species, as its root extract inhibited mushroom tyrosinase activity with 15.67 μg/mL. Therefore, further chromatographic studies were carried out on the plant extract leading to the isolation of 5 known flavonoids, namely, genistein (C1), cajanin (C2), 2'-hydoxygenistein (C3), catechin (C4), and gallocatechin (C5). Furthermore, the tyrosinase inhibitory activity of the isolated compounds was also tested, and the purified constituents showed the highest activity for C1 (31.45 μM), followed by C4 (36.86 μM), C2 (38.97 μM), C5 (60.40 μM), and C3 (69.49 μM).

Although there are a large number of *Rhynchosia* species occurring in S. Africa, research available for their biological and pharmacological importance is insufficient. For instance, there is no documented scientific or ethnomedicinal information that justifies the traditional uses of *R. villosa*. Therefore, to the best of our knowledge and according to the Sci-Finder database, the current investigation is the first scientific report to be carried out on *R. villosa*. In the present study, tyrosinase inhibitory activity demonstrated by *R. villosa* and examination of its constituents and their possible biological application in cosmetic products as skin lighteners was carried out.

To extrapolate, our research work has substantiated the projected hypotheses of the *R. villosa* potential in healing dermatological anomalies. According to our findings, it is possible that the *R.* 

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*villosa* root can bring about a transformation in the manufacturing of skin-care products by the personal care industry. Thus, its tyrosinase inhibitory effects makes it a candidate tyrosinase inhibitor that can be used as depigmenting agent for the treatment or prevention of pigmentation disorders in relation to hyperpigmentation. Moreover, obtained results justify the traditional use of *R. villosa* for skin depigmentation.

## **5.2 RECOMMENDATTIONS**

# From the above, it is suggested that:

- ✓ A reasonable amount of raw material of the examined plant species must be used in future, so to be able to obtain the available constituents in the plant of interest in adequate amounts; which in return will allow further relative biological studies on such isolated constituents.
- ✓ Appropriate action must be taken to guarantee that potential active constituents are not lost or destroyed during extraction of the plant to be examined. If the plant was chosen on the basis of traditional uses then it is needed to prepare the extract exactly as described by traditional healer in order to simulate as closely as possible the traditional herbal remedy.
- ✓ Since melanogenesis is a series of biological reactions which proceeds through a free radical mechanism, it could be disrupted by selective use of antioxidants. Hence to test the isolated constituents' antioxidant potential would do justice in confirming their uses in manufacturing of cosmetics.

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- ✓ Also, since the isolated compounds are active on the biological activity of interest, a rapid search to verify their structural integrity and binding effect to targets of the tyrosinase activity via a commonly used *in silico* or technique of structure-based virtual screening known as 'docking' may be beneficial.
- Furthermore, uses of these isolated constituents in beauty care products and pharmaceutical industry will require confirmation of their capability to reduce undesirable skin enzymatic effects in *in vivo* models. Additional investigation, for instance, cytotoxicity determination, reduction of cellular melanin formation and antityrosinase activities in B16F10 melanocytes and dermatological related clinical tests are suggested with a specific end goal to transform the present discoveries into conclusive cosmeceuticals as well as pharmaceutical product formulations.

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